

$$Q = 1 \text{ kcal} \times \frac{4186 \text{ J}}{\text{kcal}} = 4186 \text{ J}$$

$$m = 1.00 \text{ kg}$$

$$T_i = 20.0^\circ\text{C}$$

$$Q = mc\Delta T = mc(T_f - T_i)$$

$$T_f - T_i = \frac{Q}{mc}$$

$$T_f = T_i + \frac{Q}{mc}$$

a) Water

4186 J



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