

Chapter 17 Thermochemistry Final Review

1. Determine the heat required to convert 62.0 grams of ice at -10.3°C to water at 0.0°C . The specific heat capacity of ice is $2.02 \text{ J/g}^{\circ}\text{C}$.
2. Determine the energy released when converting 500.0 g of steam at 100.0°C to ice at -25.0°C .
3. Determine the final temperature when 450.2 grams of aluminum at 95.2°C is placed in an insulated calorimeter with 60.0 grams of water at 10.0°C .
4. Determine the mass of iron heated to 85.0°C to add to 54.0 grams of ice to produce water at 12.5°C . The specific heat of iron is $0.045 \text{ J/g}^{\circ}\text{C}$.

Chapter 17 Thermochemistry Text Answers

**Charles William Keenan, Donald C.
Kleinfelter, Jesse Hermon Wood**



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Early Responses to the Periodic System Masanori Kaji, Helge Kragh, Gabor Pallo, 2015-01-29 A collection of comparative studies on the reception response and appropriation of the periodic system of elements in eleven countries

The Practice of Chemistry Donald J. Wink, Sharon Fetzer-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book's unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt 832KB

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Ions in Solution J Burgess, 1999-10-01 This outline of the principles and chemical interactions in inorganic solution chemistry delivers a course module in an area of considerable complexity Problems with solutions and tutorial hints to test comprehension have been added as a feature to check readers' understanding and assist self-study Exercises and projects are also provided to help readers deepen and extend their knowledge and understanding Inorganic solution chemistry is treated thoroughly Emphasis is placed upon NMR UV VIS IR Raman spectroscopy X-ray diffraction and such topics as acid-base behaviour stability constants and kinetics The Electrical World and Engineer, 1902 *Electrical World*, 1902-07

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