

Name _____ Class _____ Date _____

Assessment

Chapter Test B

Chapter: Acid-Base Titration and pH

PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The pH scale generally ranges from
a. 0 to 1.
b. -1 to 1.
c. 0 to 7.
d. 0 to 14.
- _____ 2. During the titration of HCl with NaOH, a very rapid change in pH occurs
a. when the first addition of known solution is made.
b. when roughly equivalent amounts of H_3O^+ ions and OH^- ions become present.
c. at several points.
d. at no point.
- _____ 3. A water solution is neutral if
a. it contains no H_3O^+ ions.
b. it contains no ionized water molecules.
c. it contains no H_3O^+ ions or OH^- ions.
d. the concentrations of H_3O^+ ions and OH^- ions are equal.
- _____ 4. The antilog of a number N is
a. the inverse of N .
b. the square root of N .
c. 10 raised to the power of N .
d. N raised to the 10th power.
- _____ 5. Universal indicators
a. are mixtures of several indicator solutions.
b. are pure substances.
c. have very brief color-change intervals.
d. work well only for acidic solutions.
- _____ 6. A useful pH range for an indicator in neutralizations involving strong acids and weak bases is
a. 1.2 to 3.0.
b. 3.1 to 4.6.
c. 6.0 to 7.6.
d. 9.5 to 11.0.

Chapter 1 acid Base Titration Ph Answers

Kenneth A. Robinson



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