

Chemistry Worksheets Class 11 on Chapter 5 States of Matter- Set 1

Q-1: At 600 K, the vapour pressure of water in a closed container is 0.5 atm. If the container's volume is doubled, its vapour pressure at 600 K will be

- a) 0.2 atm
- b) 0.8 atm
- c) 0.6 atm
- d) 0.5 atm

Q-2: In the equation $PV = RT$, the value of R will depend on

- a) The nature of the gas
- b) The pressure of the gas
- c) The temperature of the gas
- d) Units of measurement

Q-3: Air at sea level is dense. This is the practical application of

- (a) Charles's law
- (b) Boyle's law
- (c) Dalton's law
- (d) Avogadro's law

Q-4: Triple point of water is

- a) 373.16 K
- b) 273.16 K
- c) 193.14 K
- d) 203.14 K

Q-5: The slope of the plot between PV and P at constant temperature is

- a) 1
- b) Zero
- c) $\frac{1}{2}$
- d) None of the above

Q-6: While pumping air into a cycle tyre, a boy noticed that volume, as well as pressure, was increasing. Does Boyle's law fail here?

Chapter 13 States Of Matter Practice Problems Answer Key

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Chapter 13 States Of Matter Practice Problems Answer Key:

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In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Chapter 13 States Of Matter Practice Problems Answer Key**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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