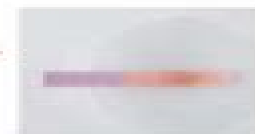


Chapter 14: Acids and Bases



Properties of Acids

- Sour taste
- React with some metals
- Turns **blue** litmus paper **red**
- React with bases



Some Common Acids

HCl, hydrochloric acid

H₂SO₄, sulfuric acid

HNO₃, nitric acid

HC₂H₃O₂, acetic acid
(common name)



What is the IUPAC name of this carboxylic acid?

Properties of Bases

- Taste bitter
- Solutions feel slippery
- Turns **red** litmus paper **blue**
- React with acids

Some Common Bases

NaOH, sodium hydroxide

KOH, potassium hydroxide

NaHCO₃, sodium bicarbonate

Arrhenius Acid-Base Theory

Arrhenius Acid - a hydrogen containing compound that ionizes to produce hydrogen ions (H⁺) when dissolved in water



Because **molecular acids are not made of ions**, they cannot dissociate.

They must be pulled apart, or **ionized**, by the water.

Chapter 14 Acids And Bases

Ronald Percy Bell



Chapter 14 Acids And Bases:

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