

# **Chemistry I Accelerated**

## **Study Guideline - Chapter 17**

### **Thermochemistry**

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By the end of this chapter the skills you should be able to demonstrate are:

1. Determine the heat of a chemical reaction in which a specified amount of substance is involved.
2. State the reason enthalpy changes in a chemical reaction.
3. Calculate enthalpies of formation.
4. Calculate changes in heat of reaction (enthalpy).
5. Describe and give examples of changes in entropy.
6. Calculate changes in entropy.
7. State two reasons reactions occur.
8. Relate Gibbs free energy to the spontaneity of reactions and to equilibrium.
9. Calculate Gibbs free energy changes involving enthalpy and entropy.

Suggested Problems: p.535-537: #56, 59, 61, 66, 68, 73, 74, 82

# Chapter 17 Thermochemistry Study Guide

**Donald J. DeCoste**



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Fundamentals of Chemistry Fred H. Redmore, 1979     *Atoms, Molecules and Reactions* Joseph D. Laposa, 1994

**CHEMISTRY** SILBERBERG, 2003     The Practice of Chemistry Donald J. Wink, Sharon Fetzer-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book's unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink [http://www.bfwpub.com/pdfs/wink/POCPowerPoint\\_Final.ppt](http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt) 832KB     **The Readers' Guide and Students' Review** Hampstead Public Libraries (London, England), 1910     *General College Chemistry* Charles William Keenan, Donald C. Kleinfelter, Jesse Hermon Wood, 1980     General Chemistry Robert K. Wismer, 1993     **Chemical principles** Paul B. Kelter, 1998     **Technical Abstract Bulletin**, 1964     Radioactive Wastes and the Ocean Paul Kilho Park, 1983 New York John Wiley and Sons 1983     **Chemistry & Chemical Reactivity** John C. Kotz, Keith F. Purcell, 1987     **Reader's Guide to Periodical Literature Supplement**, 1917

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