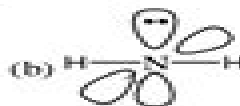




**MCEN
SECTION-A**

- The ground state term symbol for Eu^{2+} is
(a) 7F_0 (b) 7F_6 (c) 3F_0 (d) 3F_6
- Which of the following compound would be drawn most strongly into a magnetic field?
(a) TiCl_4 (b) VCl_3 (c) FeCl_2 (d) CuCl_2
- Which of the following correctly represents the balanced chemical reaction between aluminum and sulfur?
(a) $16\text{Al} + 3\text{S}_8 \longrightarrow 8\text{Al}_2\text{S}_3$ (b) $12\text{Al} + \text{S}_8 \longrightarrow 4\text{Al}_3\text{S}_2$
(c) $8\text{Al} + \text{S}_8 \longrightarrow 8\text{AlS}$ (d) $4\text{Al} + \text{S}_8 \longrightarrow 8\text{AlS}_2$
- When two ionic compounds are dissolved in water, a double replacement reaction can
(a) Never occur since all ions in water are "spectator ions"
(b) Occur if two of the ions form an insoluble ionic compound, which precipitates out of solution
(c) Occur if the ions react to form a gas, which bubbles out of the solution
(d) Occur only if the ions form covalent bonds with each other.
- Which Bronsted acid (H_2O or $\text{H}_2\text{S}_{(\text{aq})}$) is the stronger acid and why is it the stronger acid?
(a) H_2O is the stronger acid because oxygen has a greater electronegativity than sulfur, which gives the attached hydrogen atom more proton character
(b) H_2O is the stronger acid because H_2S is a gas and gases are not acids.
(c) H_2S is the stronger acid because the hydrogen-sulfur bond is much weaker than the hydrogen-oxygen bond due to a greater difference in atomic orbital energy levels.
(d) H_2S is the stronger acid because it is a heavier molecule and therefore has more energetic collisions.
- The common features among the species CN , CO , and NO are
(a) Bond order three and iso-electronic
(b) Bond order three and weak-field ligands
(c) Bond order two and stronger-field ligands
(d) Iso-electronic and weak-field ligands
- The central atom in BrF_5 has _?_ bonding pairs of electrons and _?_ non-bonding pairs of electrons
(a) 1 and 5 (b) 0 and 5 (c) 5 and 1 (d) 5 and 0
- Which of the following best represents the three-dimensional view of H_3N^+ ion?



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Yijin Wang



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