

It has a low boiling point. The electron is localized within the covalent bond of each molecule.

(b) Draw a 'dot-and-cross' diagram to show the bonding in an iodine molecule. Show only the outer shell electrons.



(c) Chlorine, bromine, iodine and astatine are all in Group VII.

(i) What is the formula for an astatide ion?

room temperature of the element

Dot And Cross Diagram For Iodine Monochloride

John Douglas Snyder

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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