

Chemistry Notes!

“Water”

1. Water is distributed on Earth as a solid, liquid and gas.

- Define the terms solute, solvent and solution.

→ **SOLUTE** is the substance (solid) that is dissolved in a solvent.
→ **SOLVENT** (liquid) is the substance that dissolves the solute.
→ **SOLUTION** is a mixture of a solid (solute) and a liquid (solvent).

- Identify the importance of water as a solvent.

→ Life developed and relies on water.
→ A solvent for all chemicals in a living cell.
→ The medium in which all the chemical reactions occur.
→ A reactant and product in many biological reactions, such as photosynthesis and cellular respiration.
→ A transport medium for carrying substances, e.g waste, nutrients and oxygen.
→ Able to dissolve numerous amounts of solutes.
→ Habitat for many species due to its specific heat capacity.
→ Used for drinking, cooking, washing and recreation.
→ For crop irrigation and farming.
→ In industry as a solvent, cleaning agent and cooling agent.

- Compare the state, percentage and distribution of water in the biosphere, lithosphere, hydrosphere and atmosphere.

Zone	State of Water	Percentage of Earth's water	Percentage water in that zone	Distribution medium
Biosphere	Liquid	0.0001%	60-90%	• In living things (cells) • photosynthesis and respiration
Hydrosphere	Liquid, solids	99.4%	99%	• Oceans, rivers, icecaps, glaciers
Lithosphere	Liquid	0.6%	10%	• Water in rocks • In soil and minerals
Atmosphere	Gas, liquid and solid	0.001%	0.5-5%	• Water vapour, rain, hail, snow

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