

a. Addition Polymer

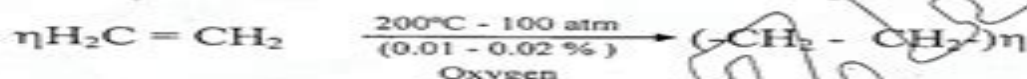
In addition polymerization, generally the unsaturated monomers combine with one another to give long chains which may be branched or un-branched straight or cyclic in nature.

Here empirical formula of monomer remains the same however the molecular mass of polymer may be several thousand a.m.u. For example the empirical formula of polythene is (CH₂ - CH₂) but molecular mass is very high.

Some common examples of addition polymers are given below.

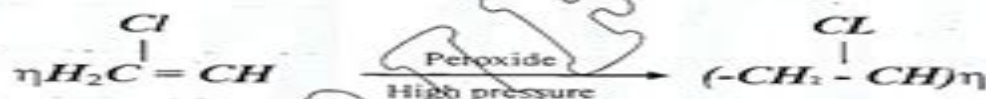
Polyethene

It is an addition polymer of ethene. It is prepared at 100 atmospheric pressure and 200°C in the presence of traces of oxygen (0.01 - 0.02 %) oxygen.



Poly Vinyl Chloride (Pvc)

Vinyl chloride, in the presence of peroxide and at high temperature, and pressure undergoes polymerization to give P.V.C.



b. Condensation Polymerization

In condensation polymerization, the monomers of two different or same compounds having different functional groups undergoes condensation with the elimination of water molecule (water) forming a long chain polymer.

Some examples of condensation polymers are given below.

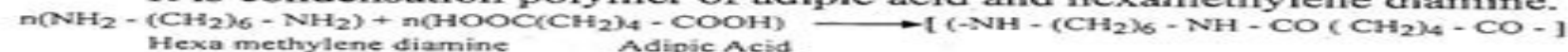
Bakelite

It is obtained by the condensation of phenol and formaldehyde (HCHO) with the elimination of water molecules.



Nylon

It is condensation polymer of adipic acid and hexamethylene diamine.



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