

# Unit 3 ~ Learning Guide Name: Aanya Nilin Shetty

## Instructions:

Using a pencil, complete the following notes as you work through the related lessons. Show ALL work as is explained in the lessons. You are required to have this package completed BEFORE you write your unit test. Do your best and ask questions if you don't understand anything!

Please print out the Formula Sheet found at the start of the course. You will need this to do assignments and tests!

\* Highly Exothermic reaction are less likely to be in equilibrium as they have a greater  $E_a$  reverse

## Equilibrium:

1. When does equilibrium occur?

Equilibrium occurs when opposing processes occur at equal rates in a closed system at a constant temperature

\* chemical equilibrium is also known as dynamic equilibrium

2. Give an example of two physical process that oppose each other during equilibrium?

A water solution in a closed container can have a physical equilibrium, and the opposing processes would be evaporation and condensation.

3. Complete the ICE chart for this reaction given that 4.0 moles of  $N_2$  gas and 8.0 moles of  $H_2$  gas are placed into a 1.00 L sealed flask and allowed to come to equilibrium. At equilibrium, 2.4 moles of  $NH_3$  are present.

|                                  | $N_{2(g)}$ | + | $3H_{2(g)}$ | $\rightleftharpoons$ | $2NH_{3(g)}$ |
|----------------------------------|------------|---|-------------|----------------------|--------------|
| Initial concentration<br>[I]     | 4.0 M      |   | 8.0 M       |                      | 0 M          |
| Change in concentration<br>[C]   | -1.2 M     |   | -3.6 M      |                      | +2.4 M       |
| Equilibrium concentration<br>[E] | 2.8 M      |   | 4.4 M       |                      | 2.4 M        |

\* Ratio of the balanced equation (1:3:2)

# Bsbfia401a Answers Learning Guide

**S Nieto**



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web 1 1 latar belakang masalah massa jenis zat atau kerapatan zat adalah salah satu sifat yang dimiliki oleh suatu zat atau benda baik itu zat padat dan zat cair massa jenis atau kerapatan zat  $\rho$  adalah massa per satuan volume pada temperatur dan tekanan tertentu dan dinyatakan sedangkan mengetahui kehidupan salah satu selama menggunakan

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web satuan massa jenis yang sering digunakan adalah  $\text{g cm}^3$  dimana  $1 \text{ g cm}^3 = 1000 \text{ kg m}^3$  walaupun zat itu beraneka ragam tetapi mempunyai beberapa sifat yang sama yaitu setiap zat menempati ruang dan mempunyai massa untuk zat padat dan zat cair mudah dibuktikan menempati ruang dan mempunyai massa

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web massa jenis yaitu perbandingan antara massa dengan volume suatu zat dan tidak dipengaruhi gaya gravitasi bumi berat jenis suatu zat berbeda dengan lainnya zat yaitu sesuatu yang memiliki massa dan memerlukan ruang berdasarkan wujudnya zat dapat dibedakan menjadi zat padat zat cair dan zat gas

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web besarnya tekanan hidrostatik tidak bergantung pada bentuk bejana dan jumlah zat cair dalam bejana tetapi tergantung pada massa jenis zat cair percepatan gravitasi bumi dan kedalamannya secara matematis tekanan hidrostatik disuatu titik misal didasar balok diturunkan dari konsep tekanan

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