

Chemistry MSL Review

Math in Chemistry

1. Give a definition and example of a qualitative measurement. Give a definition and example of a quantitative measurement.
2. How many sig figs are contained in the following numbers:
 0.067 2450 2450.0 0.0100
3. Define accuracy and precision.
4. If a block has a known density of 1.45, what is the percent error of the value 1.14 measured in lab?
5. A block of unknown composition has a mass of 23.2 grams. Using water displacement, its volume is measured at 27.9 mL. What is the block's density?
6. What is the volume of an object with a density of 1.94 g/mL and a mass of 15.5g?
7. Convert the temperature 88°C to Kelvin.
8. Metric conversions: Convert 4.58 m to cm. Convert 104.5 J to kJ. Convert 0.009 332 g to mg.
9. What is the difference between a chemical change and a physical change?

Atomic Structure

An atom is made up of protons and neutrons (both found in the nucleus) and electrons (in the surrounding electron cloud). The atomic number is equal to the number of protons. The mass number is equal to the number of protons plus neutrons. In a neutral atom, the number of protons equals the number of electrons. The charge on an ion indicates an imbalance between protons and electrons. Too many electrons produces a negative charge, too few, a positive charge.

This structure can be written as part of a chemical symbol.



1. Define the following terms:
 atomic number; mass number ; isotope ; atomic mass
2. How do you determine the number of protons in an atom? How do you determine the number of neutrons in an atom?
3. Complete the following table:

Isotope	Atomic #	# Protons	# Neutrons	Mass #
^{12}C				
	17		18	
		26		56
		17		37
			2	3
	52			128
		50	70	

- a. Consider the second and fourth row in the table above. What do they have in common?
- b. Consider the second and fourth row in the table above. What are the differences?

Dpi Study Guide Msl Chemistry

**John W. Hill, Richard Jones, Doris K.
Kolb, Terry W. McCreary, Marcia L.
Gillette**



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Yearbook of International Organizations ,1984-12 **Literature on Information Retrieval and Machine Translation** Charles F. Balz,1962 **Who's Who in the South and Southwest** Marquis Who's Who,Marquis Who's Who Staff,1998-12 Provides current coverage of a broad range of individuals from across the South and Southwest Includes approximately 17 500 names from the region embracing Alabama Arkansas Florida Georgia Kentucky Louisiana Mississippi North Carolina Oklahoma South Carolina Tennessee Texas Virginia West Virginia Puerto Rico and the Virgin Islands Because of its importance and its contiguity to the southwestern United States Mexico is also covered in this volume **Who's who**

in the West ,1999 **Study Guide for Chemical Principles [by] Steven S. Zumdahl** Steven S. Zumdahl,Paul B. Kelter,1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples **Study Guide for Chemistry** H LeMay,Theodore Brown,Bruce Bursten,Catherine Murphy,Patrick Woodward,Matthew Stoltzfus,2017-03-21 This guide assists students through the text material with chapter overviews learning objectives a review of key terms as well as self tests with answers and explanations This student guide also features MCAT practice questions **Chemistry and Student Study Guide and Study Guide, Fifth Edition and Smarthinking and Letter** Zumdahl,2002-01-01 **Student Study Guide for Chemistry** John E. McMurry,Robert C. Fay,Jill Kirsten Robinson,Julie Klare,Donnajean Fredeen,James Zubricky,2015-03-09 The Study Guide includes learning goals an overview a review section with worked examples and self tests with answers

Study Guide for Chemistry Jill Kirsten Robinson,John E. McMurry,Robert C. Fay,2019-01-04 Chemistry National Learning Corporation,2016 The College Proficiency Examination Program CPEP enables those who lack formal college course work to demonstrate their competency in various subject areas on a credit by examination basis Chemistry: Matter & Change, Study Guide For Content Mastery, Student Edition McGraw Hill,2001-03-30 Study Guide and Reinforcement Worksheets allow for differentiated instruction through a wide range of question formats There are worksheets and study tools for each section of the text that help teachers track students progress toward understanding concepts Guided Reading Activities help students identify and comprehend the important information in each chapter

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