### Bendina Basics

#### ANSWER KEY

#### Section A: Complete the chart using a periodic table to help you.

| Element   | Atomic<br>Symbol | Total # of<br>Electrons | # of Valence<br>Electrons | # of Electrons<br>Needed to<br>Gain or Lose<br>to Fill Outer Shells | Oxidation<br>Number |
|-----------|------------------|-------------------------|---------------------------|---|---------------------|
| Chlorine  | Cl               | 17                      | 7                         | Gain 1  | 1-                  |
| Potassium | к                | 19                      | 1                         | Lose I  | 1+                  |
| Magnesium | Mg               | 12                      | 2                         | Lose 2  | 2+                  |
| Fluorine  | F                | 9                       | 7                         | Gain I  | 1-                  |
| Aluminum  | Al               | 13                      | 3                         | Lose 3  | 3+                  |
| Sodium    | Na               | 11                      | 1                         | Lose 1  | 1+                  |
| Nitrogen  | N                | 14                      | 4                         | Gain 3  | 3-                  |
| Oxygen    | 0                | 8                       | 6                         | Gain 2  | 2-                  |
| Hydrogen  | н                | 1                       | 1                         | Gain or<br>Lose 1   | 1+ or 1-            |
| Carbon    | C                | 6                       | 4:                        | Gain or<br>Lose 4   | 4+ or 4-            |
| Lodine    | 1                | 53                      | 7                         | Gain 1  | 1-                  |

#### Answer these questions:

- An atom that gains one or more electrons will have a NEGATIVE charge.
- An atom that loses one or more electrons will have a POSTIVE charge.
- An atom that gains or loses one or more electrons is called an ION.
- A positive ion is called a CATION and a negative ion is called an ANION.

# **Bonding Basics Review Answer Key**

**C** Cleary

#### **Bonding Basics Review Answer Key:**

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