Name : \_\_\_\_\_\_\_ Date: \_\_\_\_\_\_

# **Classifying Chemical Reactions**

- A. Classify the following reaction as synthesis, decomposition, single-replacement, double-replacement, and combustion:
- 1) Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> + 3 BaCl<sub>2</sub> → 3 BaSO<sub>4</sub> + 2 AlCl<sub>3</sub> \_\_\_\_\_
- 2) Al<sub>2</sub>S<sub>3</sub> 2 Al + 2 S
- 3) 2 NaOH + CuSO<sub>4</sub> → Na<sub>2</sub>SO<sub>4</sub> + Cu(OH)<sub>2</sub> \_\_\_\_\_
- 4) Fe + H<sub>2</sub>SO<sub>4</sub> → FeSO<sub>4</sub> + H<sub>2</sub> \_\_\_\_\_
- 5) C<sub>4</sub>H<sub>12</sub> + 7O<sub>2</sub> 4CO<sub>2</sub> + 6H<sub>2</sub>O \_\_\_\_\_
- B. Balance the following equations and indicate the type of reaction:

1) 
$$H_2S$$
 +  $O_2$   $\longrightarrow$   $SO_2$  +  $H_2O$  \_\_\_\_\_

# **Classifying And Balancing Chemical Reactions Practice**

**M** Mark

Classifying And Balancing Chemical Reactions Practice:

# Classifying And Balancing Chemical Reactions Practice Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "Classifying And Balancing Chemical Reactions Practice," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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