

Limiting Reactant Worksheet

- 1 Given the following reaction: (First, balance the reaction)



- a Suppose 10 g of $\text{Al}_2(\text{SO}_4)_3$ reacts with 10 g of NaOH. Determine the limiting reactant.
- b Determine the number of grams of $\text{Al}(\text{OH})_3$ produced.
- c Determine the number of moles of excess reactant left over in the reaction.

- 2 Given the following reaction:



How many liters of hydrogen can be produced from the reaction between 80 g of methane and 16.3 g of water? Determine the limiting reactant first.

Cavalcade 2000 Chemistry Worksheet Limiting Reactant Worksheet

H Kauffman



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