

SECTION A

Section - A consists of Multiple Choice type questions of 1 mark each.

1. The ratio of *HCF* to *LCM* of the least composite number and the least prime number is :

- (a) 1:2
- (b) 2: 1
- (c) 1: 1
- (d) 1: 3

Answer: (a)

Explanation:

The least composite number is 4, which has prime factors 2 and 2. The least prime number is 2.

The HCF (Highest Common Factor) of 4 and 2 is 2, because that is the largest number that divides both 4 and 2.

The LCM (Least Common Multiple) of 4 and 2 is 4, because that is the smallest number that is a multiple of both 4 and 2.

Therefore, the ratio of HCF to LCM of 4 and 2 is 2:4 or 1:2.

So the answer is (a) 1:2.

2. The roots of the equation $x^2 + 3x - 10 = 0$ are :

- (a) 2, - 5
- (b) - 2, 5
- (c) 2, 5
- (d) - 2, - 5

Answer: (a)

Explanation:

$$\text{Given, } x^2 + 3x - 10 = 0$$

$$x^2 + 5x - 2x - 10 = 0$$

$$x(x + 5) - 2(x + 5) = 0$$

$$(x + 5)(x - 2) = 0$$

$x = 2, - 5$ are the roots of the given equation.

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Anthony S. Fauci



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prepare pure crystals of benzoic acid from a given impure sample

C Measurement of pH Values

- 1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper
- 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration
- 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper
- 4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH
- 5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+

D Chemical Equilibrium

- 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions
- 2 To study the shift in equilibrium between Co^{2+} and Cl^- ions by changing the concentrations of either of the ions

E Quantitative Analysis

- 1 To prepare M 10 oxalic acid solution by direct weighing method
- 2 To prepare M 10 solution of sodium carbonate by direct weighing method
- 3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid
- 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution

F Qualitative Analysis

- 1 Analysis of Anions
- 2 Analysis of Cations

G Detection of Elements in Organic Compounds

- 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test
- 2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test

INVESTIGATORY PROJECTS

A Checking of Bacterial Contamination in Water

- 1 To check the bacterial contamination in drinking water by testing sulphide ions

B Methods of Water Purification

- 1 To purify water from suspended impurities by using sedimentation
- 2 To purify water by boiling
- 3 To purify water by distillation method
- 4 To purify water by reverse osmosis technique
- 5 To purify water by GAC method
- 6 To purify water by bleach treatment
- 7 To purify water by oxidising agent
- 8 To purify water by ozone treatment method

C Water Analysis

- 1 To test the hardness of different water samples

D Foaming Capacity of Various Soaps

- 1 To compare the foaming capacity of different washing soaps
- 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap

E Tea Analysis

- 1 To study the acidity of different samples of tea leaves tea by using pH paper

F Analysis of Fruits and Vegetable Juices

- 1 To analyse the fruit and vegetable juices for the constituent present in them

G Rate of Evaporation

- 1 To study the rate of evaporation of different liquids

H Effect of Acids and Bases on Tensile Strength of Fibres

- 1 To compare the tensile strength of natural fibres and synthetic fibres
- 2 To study the effect of acids and bases on tensile strength of different fibres

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Table of Contents Cbse Class 1chemistry Question Paper 20with Solutions

1. Understanding the eBook Cbse Class 1chemistry Question Paper 20with Solutions
 - The Rise of Digital Reading Cbse Class 1chemistry Question Paper 20with Solutions
 - Advantages of eBooks Over Traditional Books
2. Identifying Cbse Class 1chemistry Question Paper 20with Solutions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Cbse Class 1chemistry Question Paper 20with Solutions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Cbse Class 1chemistry Question Paper 20with Solutions

- Personalized Recommendations
 - Cbse Class 1chemistry Question Paper 20with Solutions User Reviews and Ratings
 - Cbse Class 1chemistry Question Paper 20with Solutions and Bestseller Lists
5. Accessing Cbse Class 1chemistry Question Paper 20with Solutions Free and Paid eBooks
- Cbse Class 1chemistry Question Paper 20with Solutions Public Domain eBooks
 - Cbse Class 1chemistry Question Paper 20with Solutions eBook Subscription Services
 - Cbse Class 1chemistry Question Paper 20with Solutions Budget-Friendly Options
6. Navigating Cbse Class 1chemistry Question Paper 20with Solutions eBook Formats
- ePub, PDF, MOBI, and More
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7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Cbse Class 1chemistry Question Paper 20with Solutions
 - Highlighting and Note-Taking Cbse Class 1chemistry Question Paper 20with Solutions
 - Interactive Elements Cbse Class 1chemistry Question Paper 20with Solutions
8. Staying Engaged with Cbse Class 1chemistry Question Paper 20with Solutions
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Cbse Class 1chemistry Question Paper 20with Solutions
9. Balancing eBooks and Physical Books Cbse Class 1chemistry Question Paper 20with Solutions
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Cbse Class 1chemistry Question Paper 20with Solutions
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Cbse Class 1chemistry Question Paper 20with Solutions
- Setting Reading Goals Cbse Class 1chemistry Question Paper 20with Solutions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Cbse Class 1chemistry Question Paper 20with Solutions

- Fact-Checking eBook Content of Cbse Class 1chemistry Question Paper 20with Solutions
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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