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ACS EXAM (2024 version) Practice Questions and Answers

A 20.0 g sample of mercury(II) oxide (HgO, M = 216.6)

is heated strongly, causing it to decompose to metallic Hg and O2 gas. What volume of O2 gas is produced (measured at STP)?

- (A) 1.03 L
- (B) 2.07 L
- (C) 4.14 L
- (D) 14.0 L √√Correct-A

When 30.0 mL of 0.10 M AgNO3 is added to 30.0 mL of

formed. How much solid AgCl is produced?

0.10 M NaCl, aqueous NaNO3 and solid AgCl are

- (A) 0.0030 mol
- (B) 0.0060 mol
- (C) 0.030 mol
- (D) 0.060 mol √√Correct-A

How much Sr(OH)2 • 8 H2O (M = 265.76) is needed to

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Committee on Appropriations.
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