# Balancing Chemical Equations

Balance the equations below.

3) 
$$Z_{\text{NaCl}} + \underline{\qquad} F_2 \rightarrow \underline{\qquad} NaF + \underline{\qquad} Cl_2$$

4) 
$$2 H_2 + 0_2 \rightarrow 2 H_2 O$$

5) \_\_\_\_ Pb(OH)<sub>2</sub> + 
$$\frac{2}{2}$$
 HCl  $\rightarrow$   $\frac{2}{2}$  H<sub>2</sub>O + \_\_\_\_ PbCl<sub>2</sub>

6) 
$$2 \text{ AlBr}_3 + 7 \text{ K}_2 \text{SO}_4 \rightarrow 6 \text{ KBr} + Al_2 (\text{SO}_4)_3$$

7) \_\_\_ CH<sub>4</sub> + 
$$\frac{2}{3}$$
 O<sub>2</sub>  $\rightarrow$  \_\_\_ CO<sub>2</sub> +  $\frac{2}{3}$  H<sub>2</sub>O

8) 
$$C_0H_6 + 5_{02} \rightarrow 3_{C02} + 4_{H_20}$$

9) 
$$C_0H_{10} + 2.5O_2 \rightarrow 6.0O_2 + 9.0O_2$$

13) 
$$2 Ag_2O \rightarrow 4 Ag + 0_2$$

16) 
$$\frac{2}{K} + \frac{2}{MgBr} \rightarrow \frac{2}{KBr} + \frac{2}{MgBr}$$

17) 
$$2 \text{HCI} + 2 \text{CaCO}_3 \rightarrow \text{CaCl}_2 + \text{H}_2\text{O} + 2 \text{CO}_2$$

19) 
$$Z_{H_2O} + Z_{O_2} \rightarrow Z_{H_2O_2}$$

20) 
$$\mathbb{Z}$$
NaBr +  $\mathbb{Z}$  CaF<sub>2</sub>  $\rightarrow \mathbb{Z}$  NaF +  $\mathbb{Z}$  CaBr<sub>2</sub>

21) \_\_\_\_ H<sub>2</sub>SO<sub>4</sub> + 
$$\sum$$
 NaNO<sub>2</sub>  $\rightarrow$   $\sum$  HNO<sub>2</sub> + \_\_\_ Na<sub>2</sub>SO<sub>4</sub>

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Yan Bai

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