

CBSE Boards 2023

Answer & Solutions

Chemistry

Section-A

1. Which one of the following has lowest pK_a value?

- A. $CH_3 - COOH$
- B. $O_2N - CH_2 - COOH$
- C. $Cl - CH_2 - COOH$
- D. $HCOOH$

Answer (B)

Sol.

As, $pK_a = -\log K_a$

So, as pK_a decreases, acidity will increase, i.e. $pK_a \propto \frac{1}{\text{acidic strength}} \propto \frac{1}{-I \text{ character}}$

Among the given options, $-NO_2$ has highest $-I$ character i.e., $O_2N - CH_2 - COOH$ has highest acidic character i.e. it will have lowest pK_a value.

Hence, option (b) is correct answer.

2. Which one of the following cell was used in Apollo space program?

- A. Mercury cell
- B. MRI
- C. $H_2 - O_2$ Fuel cell
- D. Dry cell

Answer (C)

Sol.

One of the most successful fuel cell uses the reaction of hydrogen with oxygen to form water. This fuel cell was used for providing electric power in Apollo space program.

3. Consider the following standard electrode potential values:



What is the cell potential for the redox reaction?

- A. -2.28 V
- B. -0.74 V
- C. $+0.74 \text{ V}$
- D. $+2.28 \text{ V}$

Answer (C)

Sol.



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David Kirk



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