#### PHYSICAL PHARMACEUTICS

#### QUESTIONS:

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|-----|-------------------|-----|------------------|-------------------|----------------------|------|
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- a) Mass. b) Volume. c) Temperature. d)Pressure.
- When a large amount of solute dissolves in a given amount of solvent it forms:
  - a) Delute solution.
  - b) Saturated solution.
  - c) Aqueous solution.
  - d) Concentrated Solution.
- Solution which can hold no more solute is called:
  - a) Delute solution.
  - b) Saturated solution.
  - d) Aqueous solution.
  - d) Concentrated Solution.
- 4. Which is not a colligative property:
  - a) Roult's law.
  - b) Depression in freezing point.
  - d) Osmotic pressure.
  - d) Melangpoint.
- Azeotropic mixtures are:
  - a) Constant temperature boiling mixtures.
  - Those which boils at different temperatures.
  - c) Mixture of two solids.
  - d) None of above.
- The pH of pharmaceutical buffer system can be calculated by:
  - a) pH partition theory.
  - b) Nopes Whitney law.
  - d) Henderson Hasselbaltch equation.
  - d) Michaelia Menten equation.
- 7. Unit of surface tension is:
  - a) Dyney meter
  - b) Dyne/cm
  - d) Calam
  - d) None of the above.
- S. Two solutions are said to be isotonic if they exert same......
  - a) Viscosity.

- c) O smotic pressure.
- b) Surface tension.
- d) None of the above.
- 9. Dissolution is affected by:
  - a) Surface area.
  - b) Viscosity.
  - c) Temperature.
  - d) All of the above.

# **Basic Physical Pharmacy Questions With Answer**

Joseph K. H. Ma,Boka Hadzija

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