## Science

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(Chapter - 3) (Atoms and Molecules) (Exercises)

(Class - IX)

#### Question 1:

A 0.24 g sample of compound of oxygen and boron was found by analysis to contain 0.096 g of boron and 0.144 g of oxygen. Calculate the percentage composition of the compound by weight.

#### **Z**Answer 1:

Mass of boron = 0.096g

(Given) Mass of oxygen = 0.144g (Given) Mass of sample = 0.24g (Given)

Thus, percentage of boron by weight in the compound  $=\frac{0.096 \times 100}{0.24}\% = 40\%$ 

Thus, percentage of boron by weight in the compound  $= \frac{0.144 \times 100}{0.24} \% = 60\%$ 

#### Question 2:

When 3.0 g of carbon is burnt in 8.00 g oxygen, 11.00 g of carbon dioxide is produced. What mass of carbon dioxide will be formed when 3.00 g of carbon is burnt in 50.00 g of oxygen? Which law of chemical combination will govern your answer?

#### **\_Answer 2**:

Carbon + Oxygen → Carbon dioxide 3g of carbon reacts with 8 g of oxygen to produce 11g of carbon dioxide. If 3g of carbon is burnt in 50g of oxygen, then 3g of carbon will react with 8 g of oxygen. The remaining 42 g of oxygen will be left un-reactive. In this case also, only 11g of carbon dioxide will be formed. The above answer is governed by the law of constant proportions.

#### Question 3:

What are polyatomic ions? Give examples.

#### **Z**Answer 3:

A polyatomic ion is a group of atoms carrying a charge (positive or negative). For example, ammonium ion  $(NH_s^+)$ , hydroxide ion  $(OH^-)$ , carbonate ion  $(CO_s^{2-})$ , sulphateion  $(CO_s^{2-})$ .

#### Question 4:

Write the chemical formulae of the following.

- (a) Magnesium chloride
- (b) Calcium oxide
- (c) Copper nitrate
- (d) Aluminium chloride
- (e) Calcium carbonate.

#### Answer 4:

- (a) Magnesium chloride →MgCl<sub>2</sub>
- (b) Calcium oxide →CaO
- (c) Copper nitrate →Cu(NO<sub>3</sub>)<sub>2</sub>
- (d) Aluminium chloride →AlCl<sub>3</sub>
- (e) Calcium carbonate →CaCO<sub>3</sub>

#### Question 5:

Write the chemical formulae of the following,

- (a) Quick lime
- (b) Hydrogen bromide
- (c) Baking powder
- (d) Potassium sulphate.

# **Bc Science 9 Chapter 3 Test**

**Brendan G. Carr** 

#### **Bc Science 9 Chapter 3 Test:**

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