



I Semester B.C.A. Degree Examination, Nov./Dec. 2017
(2014-15 and Onwards) (F + R) (CBCS)
BCA – 105 T : DISCRETE MATHEMATICS

Time : 3 Hours

Max. Marks : 100

Instruction : Answer all Sections.

SECTION – A

I. Answer any ten of the following :

(10×2=20)

1) If $A = \{2, 3, 4, 5\}$ and $B = \{0, 1, 2, 3\}$, find $A \cap B$.

2) If $A = \{x^2 - 5x + 6 = 0, x \in \mathbb{N}\}$ and $B = \{3, 4, 5\}$, find $A \times B$.

3) Define contradiction.

4) Define unit matrix with example.

5) If $A = \begin{bmatrix} 3 & 2 \\ -1 & 4 \end{bmatrix}$ and $B = \begin{bmatrix} 3 & 5 \\ -2 & 4 \end{bmatrix}$, find $2A + 3B$.

6) Find the characteristic roots of the matrix $A = \begin{bmatrix} 3 & 0 \\ 2 & 5 \end{bmatrix}$.

7) Prove that $\log_{3a} 2a \cdot \log_{4a} 3a = \frac{1}{2}$.

8) If ${}^nC_{30} = {}^nC_5$, find 'n'.

9) Define group.

10) If $\vec{a} = 2\hat{i} + 3\hat{j} + 4\hat{k}$, $\vec{b} = \hat{i} - 2\hat{j} + \hat{k}$, find $|2\vec{a} + \vec{b}|$.

11) Find the distance between the points A(2, -3) and B(4, 5).

12) Write the slope of the line $4x - 3y + 2 = 0$.

Bca 1st Year Question Papers 105

**Lakshman Prasad,Deepak
Paliwal,Mansi Garg,Neha Minglani
Sachdeva,Sanubia Saleem**

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alcohols ethanol 2 methylpropan-2-ol butan-2-ol but-2-en-1-ol $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$ ii $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$ iii $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$ iv $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$

aldehydes and ketones draw the structures for the following compounds ethanal

butanone also known as butan-2-one but the 2 is unnecessary and is often left out 2-methylpropanal

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question 1 basic hydrocarbon nomenclature part a enter the number of carbons indicated by each of the following root names in the designated answer box part b the line formula for a branched alkane is shown below part c the line formula for another branched alkane is

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solutions 1 aromatic compound iodobenzene 2 alkane 3 methylhexane 3 ketone 4 alkane alkyl halide 3 chloro 4 ethyl 2 4 dimethyloctane 5 aromatic compound o diethylbenzene or ortho diethylbenzene 6 alkene 3 methylpentene 7 ester 8 aromatic compound p bromotoluene or para bromotoluene 9 alkyne

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ethoxy ethane methyl ethane ethyl ethane

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