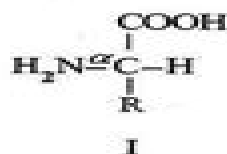


31. (i) Assign D – and L– configuration of  $\alpha$  – amine acids for the following structures I and II.



- (ii) On electrolysis in acidic medium amino acids migrate towards the cathode while in alkaline medium they migrate towards anode. Explain. 1+2 = 3  
*Question Nos. 32-34 are Essay (E) type of 5 marks each.*

32. (a) Define colligative properties.  
 (b) Establish the relationship between the relative lowering of vapour pressure of a solution and mole fraction of the solute in it when the solvent alone is volatile.  
 (c) The van't Hoff factor ( $i$ ) of a solution is more than one. What does it indicate?

1+3+1= 5

33. (i) Why are halogens placed in Group 17 ?  
 (ii) Halogens except fluorine exhibit higher oxidation state. Explain why.  
 (iii) Why are boiling points of noble gases very low? How the boiling points vary on going down the group (gr-18) ?

1+2+2=5

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**Buchanan**



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