

Name: \_\_\_\_\_

Date: \_\_\_\_\_

## Student Exploration: Covalent Bonds

**Vocabulary:** covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers?

Flora can give Frank her blue marker, and in return, Frank can give Flora his red marker. This way, both will have a full set of eight markers. \_\_\_\_\_

2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?


Otto can give Olivia his black and brown markers, and Olivia can give Otto her purple and green markers. This exchange would help both of them complete their sets with all six markers. \_\_\_\_\_

### Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have?
2. Click **Pause** (). Drag a valence electron from the left atom to the right atom. Click **Play**.

What happens? \_\_\_\_\_

# Covalent Bond Gizmo Worksheet Answers

**Henry Sinclair Pickering**



**Covalent Bond Gizmo Worksheet Answers:**

**The Covalent Bond** Henry Sinclair Pickering, 1977

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