



# Determining Empirical Formula Worksheet



1. A solvent is found to be 50.0% oxygen, 37.5% carbon, and 12.5% hydrogen. What is the empirical formula of this solvent?
2. A particular sugar is determined to have the following composition: 40.0% carbon, 6.7% hydrogen, and 53.5% oxygen. Determine the empirical formula of this sugar molecule.
3. If 4.04 g of nitrogen combine with 11.46 g of oxygen to produce a compound with a molar mass of 108.0 g, what is the molecular formula of this compound?
4. The molar mass of a compound is 92 g. Analysis of the sample indicates that it contains 0.606 g N and 1.390 g O. Find the compound's molecular formula.

# Determining Empirical Formulas Answer Key

**Donald J. Wink, Sharon Fetzer-  
Gislason, Sheila McNicholas**



## Determining Empirical Formulas Answer Key:

The Practice of Chemistry Donald J. Wink, Sharon Fetzer-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math. The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving. The book's unique focus on actual chemical practice, extensive study tools, and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors. This special PowerPoint tour of the text was created by Don Wink. [http://www.bfwpub.com/pdfs/wink/POCPowerPoint\\_Final.ppt](http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt) 832KB

Chemical Reactions Mr. Ajeet Singh, 2018-05-22 A fundamental text covering types of chemical reactions, reaction energetics, mechanisms, and industrial applications.

General Chemistry Donald A. McQuarrie, Stanley Gill, 2011-06-15 This Fourth Edition of McQuarrie's classic text offers a thorough revision and a quantum leap forward from the previous edition. Taking an atoms first approach, it promises to be another ground-breaking text in the tradition of McQuarrie's many previous works. This outstanding new text, available in a soft cover edition, offers professors a fresh choice and outstanding value.

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Study Guide for Chemical Principles Thomas Elliott Taylor, 1979

Prentice Hall Chemistry, 2000

**Chemistry Homework** Frank Schaffer Publications, Joan DiStasio, 1996-03 Includes the periodic table, writing formulas, balancing equations, stoichiometry.

problems and more     Introduction to Chemical Principles Howard Stephen Stoker,1983     **Introduction to Chemistry, Laboratory Manual** T. R. Dickson,1994-12-23 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and a chemical vision learning problem solving methods and utilizing group work and discussion activities These themes involve and engage the students in their own learning processes they are challenged to be active The presentation of topics has been altered to include a new chapter which introduces the students to scientific thinking and shows that chemistry involves interesting and relevant topics The reorganization presents many core concepts in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the chapters referring to health technology the environment and society as well as to specific tools of direct use to students     *Merrill Chemistry* Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998     Addison-Wesley Chemistry ,1990     *Holt Chemistry* Ralph Thomas Myers,2004     Preparatory Chemistry Howard Stephen Stoker,1985     *Basic Chemistry* William S. Seese,Guido H. Daub,1977     **Students Study Guide** Fredeen,Pearson,2003-07     **Alchemy Revisited** C. G. Vlassis,1976     Organic Chemistry G. Marc Loudon,1988

## **Determining Empirical Formulas Answer Key** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Determining Empirical Formulas Answer Key**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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