

Date Performed: 9/18/06
Date Submitted: 9/25/06
Instructor: Miao

Name: Jon Conlon
Partner: Jakob Vasiloff

Determination of a Chemical Formula

Objective:

To determine the percentage composition and simplest formula for an oxide of copper by its reduction with methane gas (CH_4) at approximately 500°C . This experiment is a study of stoichiometry.

Experimental Data:

Mass of Test Tube	$17.726 \pm .001 \text{ g}$
Mass of Test Tube and Cu_2O	$19.705 \pm .001 \text{ g}$
Mass of Cu_2O	$1.979 \pm .001 \text{ g}$
Mass of Test Tube and Cu	$19.313 \pm .001 \text{ g}$
Mass of Cu	$1.587 \pm .001 \text{ g}$
Mass of O	$0.392 \pm .001 \text{ g}$

Sample Calculation:

Mass percentage calculation:

$$\frac{\text{Mass of Atom}}{\text{Mass of Compound}} \times 100 = \frac{1.587 \pm .001 \text{ g of Cu}}{1.979 \pm .001 \text{ g of Cu}_2\text{O}} \times 100 = 79.73\%$$

Calculation of a Chemical Formula:

Assume a 100 g compound and calculate individual mass using mass percentage.

$$79.73 \text{ g of Cu} \times \frac{1 \text{ mol of Cu}}{63.55 \text{ g per mol}} = 1.255 \text{ mol of Cu}$$

$$20.07 \text{ g of O} \times \frac{1 \text{ mol of O}}{16.00 \text{ g per mol}} = 1.254 \text{ mol of O}$$

Results and Conclusions:

Mass of Test Tube	$17.726 \pm .001 \text{ g}$
Mass of Test Tube and Cu_2O	$19.705 \pm .001 \text{ g}$
Mass of Cu_2O	$1.979 \pm .001 \text{ g}$
Mass of Test Tube and Cu	$19.313 \pm .001 \text{ g}$
Mass of Cu	$1.587 \pm .001 \text{ g}$
Mass of O	$0.392 \pm .001 \text{ g}$
Mass Percentage of Cu	79.73%
Mass Percentage of O	20.07%

Cu and O have a one to one mole ratio; therefore the empirical formula is CuO .

Determination Of Chemical Formula Lab Report

Chunlong Zhang



Determination Of Chemical Formula Lab Report:

English in Analytical Chemistry. Communicating about Methods & Techniques. Книга для студента Надежда Зорина, Александра Соболева, 2022-10-19 **Fundamentals of Environmental Sampling and Analysis** Chunlong Zhang, 2007-03-09 An integrated approach to understanding the principles of sampling chemical analysis and instrumentation This unique reference focuses on the overall framework and why various methodologies are used in environmental sampling and analysis An understanding of the underlying theories and principles empowers environmental professionals to select and adapt the proper sampling and analytical protocols for specific contaminants as well as for specific project applications Covering both field sampling and laboratory analysis Fundamentals of Environmental Sampling and Analysis includes A review of the basic analytical and organic chemistry statistics hydrogeology and environmental regulations relevant to sampling and analysis An overview of the fundamentals of environmental sampling design sampling techniques and quality assurance quality control QA QC essential to acquire quality environmental data A detailed discussion of the theories of absorption spectroscopy for qualitative and quantitative environmental analysis metal analysis using various atomic absorption and emission spectrometric methods and the instrumental principles of common chromatographic and electrochemical methods An introduction to advanced analytical techniques including various hyphenated mass spectrometries and nuclear magnetic resonance spectroscopy With real life case studies that illustrate the principles plus problems and questions at the end of each chapter to solidify understanding this is a practical hands on reference for practitioners and a great textbook for upper level undergraduates and graduate students in environmental science and engineering **Scientific and Technical Aerospace Reports** ,1969 *Report of the Division of Chemistry* Canada. Department of Agriculture. Division of Chemistry, 1924 Chemistry Gary S. Thorpe, 2001 CliffsAP study guides help you gain an edge on Advanced Placement exams Review exercises realistic practice exams and effective test taking strategies are the key to calmer nerves and higher APa scores CliffsAP Chemistry is for students who are enrolled in AP Chemistry or who are preparing for the Advanced Placement Examination in Chemistry Inside you ll find hints for answering the essay and multiple choice sections a clear explanation of the exam format reviews of all 22 required labs a look at how exams are graded and more Realistic full length practice exam Answers to commonly asked questions about the AP Chemistry exam Study strategies to help you prepare Thorough review of the key topics that are sure to be on the test Sample laboratory write ups The AP Chemistry exam is coming up Your thorough understanding of months and months of college level chemistry coursework is about to be evaluated in a 3 hour examination CliffsAP Chemistry includes the following material to you do the very best job possible on the big test Gravimetrics Electronic structure of atoms Covalent bonding and ionic bonding Acids and bases Reduction and oxidation Organic chemistry and nuclear chemistry Writing and predicting chemical reactions This comprehensive guide offers a thorough review of key concepts and detailed answer explanations It s all you

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