

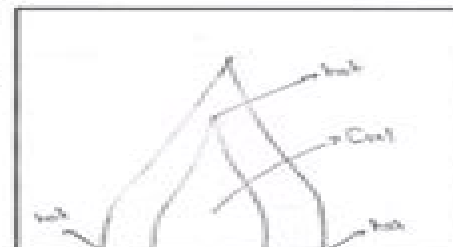
Experiment 1 Report Sheet

Basic Laboratory Operations

A. Bunsen Burner

1. Instructor's Approval of a well-adjusted Bunsen flame: _____
2. a. At right, draw a sketch of the heat zones for a nonluminous flame as directed with the wire gauze positioned parallel to the burner barrel. Label the "cool" and "hot" zones.
 - b. What happens to the "cool" and "hot" zones of a nonluminous flame, when the air-control valve is closed (nearly closed)? Explain.

As the air-control valve approaches the nearly closed position, the "cool" and "hot" zones merge into one flame.



B. Laboratory Balances

1. Determine the mass of the following objects on your assigned balance. Express your results with the correct sensitivity.

Balance no. _____ Sensitivity 0.001 g

Object	Description of Object (size or volume)	Mass (g)
Test tube		<u>5.240 g</u>
Beaker	<u>50 mL</u>	<u>29.052 g</u>
Spatula		<u>7.401 g</u>
Graduated cylinder	<u>10 mL</u>	<u>15.711 g</u>
Other (see instructor)	<u>Unknown #2</u>	<u>2.135 g</u>

2. Precision of Measurement. Balance no. _____

Mass of graduated cylinder (if not tared) 15.711 g

	Trial 1	Trial 2	Trial 3	Trial 4	Trial 5	Trial 6
a. Mass of cylinder + H ₂ O (l) (g)	<u>20.261 g</u>	<u>25.423 g</u>	<u>21.002 g</u>			
b. Mass of 7 mL H ₂ O (l) (g)	<u>4.55 g</u>	<u>4.71 g</u>	<u>4.511 g</u>			
c. Average mass of 7 mL H ₂ O (l) (g)			<u>4.59 g</u>			

Comment on the precision of the mass measurements for the water samples.

Trial #3 does not seem to be precise.

Date Analysis, B

Beran Laboratory Manual Experiment 2 Report Sheet

A Loxley



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