

Name \_\_\_\_\_ Date \_\_\_\_\_ Per \_\_\_\_\_

**WORKSHEET: Chemical Bonding – Ionic & Covalent!**
**REMEMBER...**
**Ionic Bond**

between a Metal and Non-Metal

(M + NM)

**Covalent Bond**

between a Non-Metal and Non-Metal

(NM + NM)

**PART 1:** Determine if the elements in the following compounds are metals or non-metals. Describe the type of bonding that occurs in the compound.

Compound	Element 1 (metal or non-metal?)	Element 2 (metal or non-metal?)	Bond Type
NO <sub>2</sub>	N = non-metal	O = non-metal	covalent
NaCl			
SO <sub>2</sub>			
PI <sub>3</sub>			
MgBr <sub>2</sub>			
CaO			
H <sub>2</sub> O			
K <sub>2</sub> O			
AlF <sub>3</sub>			
O <sub>2</sub>			
CuCl <sub>2</sub>			
NO <sub>2</sub>			
CO <sub>2</sub>			
HF			
Rb <sub>2</sub> S			
NBr <sub>3</sub>			
Fe <sub>2</sub> O <sub>3</sub>			
CCl <sub>4</sub>			

# Bond Identification Answers

**E Durkheim**



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## **Table of Contents Bond Identification Answers**

1. Understanding the eBook Bond Identification Answers
  - The Rise of Digital Reading Bond Identification Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Bond Identification Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Bond Identification Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Bond Identification Answers
  - Personalized Recommendations
  - Bond Identification Answers User Reviews and Ratings
  - Bond Identification Answers and Bestseller Lists
5. Accessing Bond Identification Answers Free and Paid eBooks
  - Bond Identification Answers Public Domain eBooks
  - Bond Identification Answers eBook Subscription Services
  - Bond Identification Answers Budget-Friendly Options
6. Navigating Bond Identification Answers eBook Formats



- ePub, PDF, MOBI, and More
- Bond Identification Answers Compatibility with Devices
- Bond Identification Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Bond Identification Answers
  - Highlighting and Note-Taking Bond Identification Answers
  - Interactive Elements Bond Identification Answers
- 8. Staying Engaged with Bond Identification Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Bond Identification Answers
- 9. Balancing eBooks and Physical Books Bond Identification Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Bond Identification Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Bond Identification Answers
  - Setting Reading Goals Bond Identification Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Bond Identification Answers
  - Fact-Checking eBook Content of Bond Identification Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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