

Unit Test: Chemical Bonding (A) **ANSWER KEY**

100

**Part I: Matching** (2 points each) Please match the term with the correct definition. There will be extra choices and some choices may be used more than once.

- |                      |                         |                      |
|----------------------|-------------------------|----------------------|
| A. Electronegativity | F. Molecular Formula    | K. Crystal           |
| B. Covalent bonding  | G. Linear               | L. Pyramid           |
| C. Ductile           | H. Non-polar            | M. Polar             |
| D. Aqueous           | I. Solid                | N. VSEPR             |
| E. Tetrahedral       | J. Trigonal bipyramidal | O. Malleable         |
| P. Octet Rule        | Q. Ionic bonding        | R. Empirical Formula |

1.   K   A 3-D, repeating pattern of + and - ions, formed by ionic compound
2.   A   Tendency for an atom to attract electrons when chemically bonded to another atom
3.   B   A sharing of electrons
4.   P   Atoms will gain or lose enough electrons in order to become isoelectronic with a noble gas
5.   Q   A transfer of electrons
6.   R   A chemical formula that is arranged in the smallest whole number ratio
7.   D   The term that means dissolved in water
8.   F   A chemical formula that describes the makeup of a single molecule
9.   J   The shape (geometry) that is always an exception to the octet rule
10.   M   A bond where electrons are shared unequally between atoms
11.   L   One of the shapes (geometries) that will always be polar
12.   H   A bond where electrons are shared equally between atoms

**Part II: Multiple Choice** (2 points each) Please circle the most correct answer. Write your answer on the given lines.

1.   E   A triple covalent bond shares \_\_\_\_\_ electrons between atoms.  
a. two  
b. three  
c. four  
d. five  
e. six  
f. Unable to tell
2.   C   Which of the following sets of atoms can be joined by a covalent bond?  
a. Be and F  
b. Na and Cl  
c. S and F  
d. Mg and Na  
e. Li and Cl  
f. None of these

# Bonding Test Answer Key

**S Baum**



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*Structure, Bonding and Main Group Chemistry* Rod Beavon, Alan Jarvis, 2003 The revised edition of the highly successful Nelson Advanced Science series for A Level Chemistry Structure Bonding and Main Group Chemistry provides full content coverage of Unit 1 of the AS and A2 specifications *CCEA GCSE Double Award Science* Denmour Boyd, Nora Henry, Frank McCauley, Alyn G. McFarland, James Napier, Roy White, 2017-12-18 Exam Board CCEA Level GCSE Subject Science First Teaching September 2017 First Exam June 2019 Build your students scientific thinking and practical skills with this textbook developed specifically for the 2017 GCSE specifications from the No 1 publisher for CCEA GCSE Science Develop understanding with clear Examples Tips and Practical activities Prepare students for assessment with Test Yourself questions Maths practice and Exam style questions throughout Supports Foundation and Higher tier students in one book

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**AERA.**, 1926 South Western Reporter. Second Series, 1928 *Patents Abstracts of Japan*, 1982 *The South Western Reporter*, 1916 Includes the decisions of the Supreme Courts of Missouri Arkansas Tennessee and Texas and Court of Appeals of Kentucky Aug Dec 1886 May Aug 1892 Court of Appeals of Texas Aug 1892 Feb 1893 Jan Feb 1928 Courts of

Civil and Criminal Appeals of Texas Apr June 1896 Aug Nov 1907 Court of Appeals of Indian Territory May June 1927 Jan Feb 1928 Courts of Appeals of Missouri and Commission of Appeals of Texas     *The Southwestern Reporter* ,1920     Thermal Spray Manual Ray Travis,1996     Aeronautical Engineering Review ,1957     **Southern Pulp and Paper Journal** ,1967-07

**Holt Chemistry** Ralph Thomas Myers,2004     **Corrosion Engineering** ,2002     Student Study Guide to Accompany Human Biology Sylvia S. Mader,2003-08 This best selling text emphasizes the relationship between humans and other living things Intended for an introductory course this text provides students with a firm grasp of how their bodies function and how the human population can become more fully integrated into the biosphere An Online Learning Center tied directly to the text via icons will direct students to activities or animations that gives a visual example of difficult processes as well as Working Together boxes to emphasize homeostasis



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