

## Bonding Basics

## ANSWER KEY

Section A: Complete the chart using a periodic table to help you.

Element	Atomic Symbol	Total # of Electrons	# of Valence Electrons	# of Electrons Needed to Gain or Lose (to Full Outer Shell)	Oxidation Number
Chlorine	Cl	17	7	Gain 1	1-
Potassium	K	19	1	Lose 1	1+
Magnesium	Mg	12	2	Lose 2	2+
Fluorine	F	9	7	Gain 1	1-
Aluminum	Al	13	3	Lose 3	3+
Sodium	Na	11	1	Lose 1	1+
Nitrogen	N	14	4	Gain 3	3-
Oxygen	O	8	6	Gain 2	2-
Hydrogen	H	1	1	Gain or Lose 1	1+ or 1-
Carbon	C	6	4	Gain or Lose 4	4+ or 4-
Iodine	I	53	7	Gain 1	1-

Answer these questions:

- An atom that gains one or more electrons will have a **NEGATIVE** charge.
- An atom that loses one or more electrons will have a **POSTIVE** charge.
- An atom that gains or loses one or more electrons is called an **ION**.
- A positive ion is called a **CATION** and a negative ion is called an **ANION**.

# Bonding Basics Section A Answer Key

**Arshad Iqbal**



## **Bonding Basics Section A Answer Key:**

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In some sort of eaten by monitors and the ceaseless chatter of immediate communication, the melodic elegance and mental symphony produced by the prepared term frequently diminish into the backdrop, eclipsed by the relentless sound and disruptions that permeate our lives. However, nestled within the pages of **Bonding Basics Section A Answer Key** an enchanting fictional value full of natural feelings, lies an immersive symphony waiting to be embraced. Constructed by an elegant musician of language, that charming masterpiece conducts visitors on an emotional journey, skillfully unraveling the concealed melodies and profound impact resonating within each cautiously constructed phrase. Within the depths of this poignant examination, we shall examine the book is key harmonies, analyze their enthralling writing style, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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