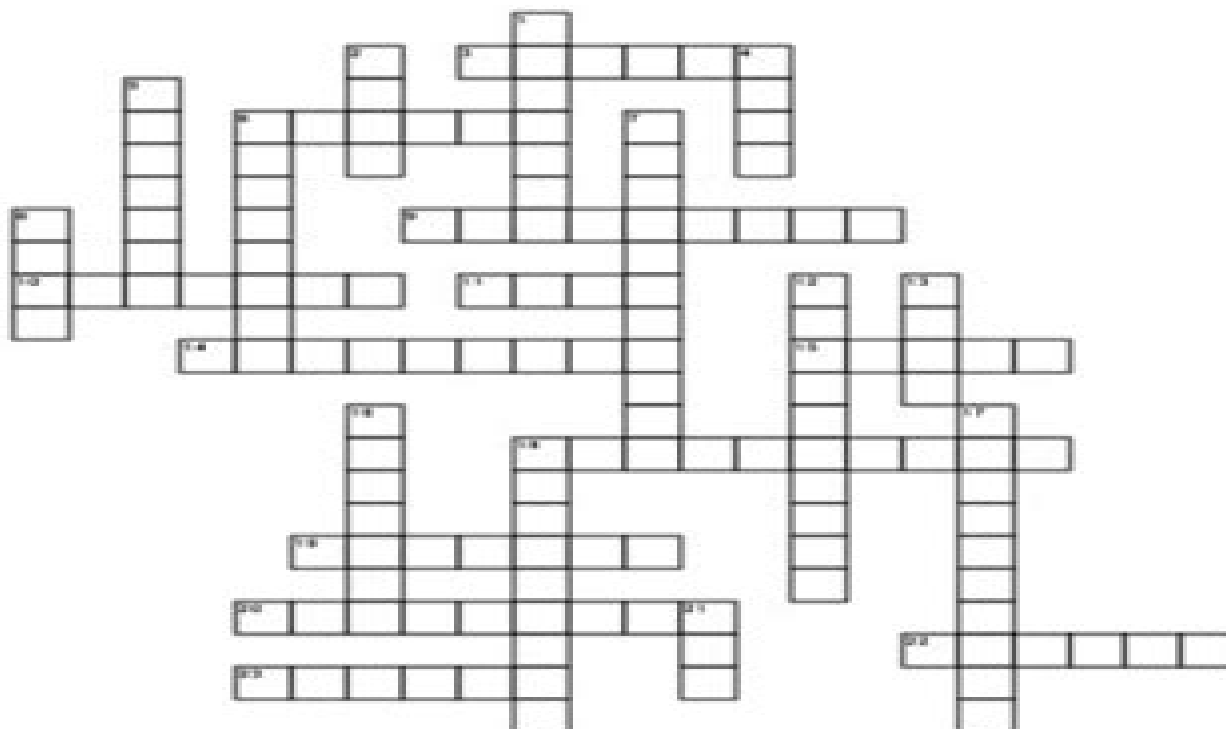


Acids, Bases and Solutions



Across

- [3] _____ paper changes from red to blue.
 [5] The substance that is dissolved in the solvent.
 [9] Acids are described as _____, meaning they wear away other materials.
 [10] A substance that takes in or dissolves another substance.
 [11] A substance with a pH greater than 7 is a/an _____.
 [14] A base produces _____ ions (OH^-) in water.
 [15] Solutes _____ the boiling point of a solution above that of the solvent.
 [18] A _____ is a mixture in which particles can be seen and easily separated by settling or filtration.
 [19] Ionic compounds in water _____ electric current.
 [20] An _____ changes color depending on the solution's pH.
 [22] The taste of a basic solution.
 [23] A solution that contains very little solute is a/an _____ solution.

Down

- [1] An example of an acidic solution.
 [2] When an acid and a base react, they form water and a _____.
 [4] An acidic solution tastes _____.
 [5] If something has a pH of 7.0 it is _____.
 [6] Bases feel _____ because they react with oils in your skin to form soap.
 [7] Solution is another name for a _____ mixture.
 [8] Unlike most solids, gases become _____ soluble when the temperature goes up.
 [12] Acids react with _____ to produce carbon dioxide gas.
 [13] A substance that gives up hydrogen ions (H^+) in a water solution.
 [16] A _____ is a mixture containing small, undissolved particles that do not settle out.
 [17] _____ is a measure of how much solute can dissolve in a solvent at a given temperature.
 [18] A solution in which no more solute will dissolve.
 [21] An acidic solution turns litmus _____.

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RD Boyd



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