Matter Homework Packet

Name	
	Period

Physical and Chemical Changes and Properties of Matter Worksheet

Classify the following as chemical change (cc), chemical property (cp), physical change (pc), or physical property (pp).

1.	Is is	Heat conductivity	8.	CP	Combustible
2.	CC	Silver tarnishing	9.	PC	Water freezing
3.	PC	sublimation	10.	CC	Wood burning
4.	PC	magnetizing steel	1.1.	CP	Acid resistance
5.	PP	length of metal object	12.	PP	Brittleness
6.	PC	shortening melting	1.3.	CC	Milk souring
7.	CC	exploding dynamite	1.4.	CC	baking bread

Identify the following as being true or false to the left of the sentence.

- T 15. A change in size or shape is a physical change.
 - T 16. A chemical change means a new substance with new properties was formed.
- F 17. An example of a chemical change is when water freezes.
 T 18. When platinum is heated, then cooled to its original state, we say this is a physical change.
- F 19. When milk turns sour, this is a physical change because a change in odor does not indicate a chemical change.
- temperature decreases. This must be a chemical change.

Identify each of the following as a physical or chemical change.

- You leave your bicycle out in the rain and it rusts.
- PC A sugar cube dissolves.
- Scientist break-up water into oxygen and hydrogen gas.
- CC Burning coal for a barbecue.
- PC Trimming a bush because it has grown too tall.

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The Code of Civil Procedure of the State of California California, Nathan Newmark, 1880 The Code of Civil Procedure of the State of California, Adopted March 11th, 1872, and Amended in 1883 California, 1883 The Code of Civil Procedure of the State of California, Adopted March 11th, 1872, and Amended in 1880 California, 1880 The Code of Civil Procedure of the State of California, Adopted March 11th, 1872, and Amended in 1881 California, Nathan Memoranda Chiefly Relating to the Classification and Incidence of Imperial and Local Taxes ... Great Newmark, 1880 **Introduction to Chemistry** Britain. Royal Commission on Local Taxation, 1899 Matter Prentice-Hall Staff, 1994 Richard C. Bauer, James P. Birk, Pamela Marks, 2024 The sixth edition of Introduction to Chemistry continues to build on our belief that students learn best when the text and our classroom presentations focus on a conceptual approach to chemistry Our class meetings are significantly different from traditional lecture presentations in many ways Beginning with the first week of classes and continuing through the rest of the semester we follow a sequence of topics that allows us to explain macroscopic phenomena from a molecular perspective This approach places emphasis on conceptual understanding over algorithmic problem solving **Prentice Hall Physical Science Concepts in Action Program Planner National**

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Railway Age ,1940

Matter, Building Block of the Universe ,1993 Agricultural Education ,1972 The Postal Record ,1923
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