

Electrons in Atoms

Section 5.1 Light and Quantized Energy

In your textbook, read about the wave nature of light.

Use each of the terms below just once to complete the passage.

amplitude
light

energy
wave

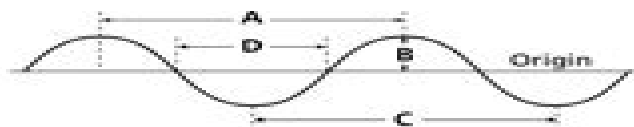
frequency
wavelength

hertz
speed

Electromagnetic radiation is a kind of **(1)** _____ that behaves like a(n) **(2)** _____ as it travels through space. **(3)** _____ is one type of electromagnetic radiation. Other examples include X rays, radio waves, and microwaves.

All waves can be characterized by their wavelength, amplitude, frequency, and **(4)** _____. The shortest distance between equivalent points on a continuous wave is called a(n) **(5)** _____. The height of a wave from the origin to a crest or from the origin to a trough is the **(6)** _____. **(7)** _____ is the number of waves that pass a given point in one second. The SI unit for frequency is the **(8)** _____, which is equivalent to one wave per second.

Use the figure to answer the following questions.



9. Which letter(s) represent one wavelength?

10. Which letter(s) represent the amplitude? _____
11. If twice the length of A passes a stationary point every second, what is the frequency of the wave?

Chapter Supplemental Problems Electrons In Atoms

Answer Key

Michael Browne



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CliffsStudySolver: Chemistry Charles Henrickson, 2007-05-03 The CliffsStudySolver workbooks combine 20 percent review material with 80 percent practice problems and the answers to help make your lessons stick CliffsStudySolver Chemistry is for students who want to reinforce their knowledge with a learn by doing approach Inside you ll get the practice you need to learn Chemistry with problem solving tools such as Clear concise reviews of every topic Practice problems in every chapter with explanations and solutions A diagnostic pretest to assess your current skills A full length exam that adapts to your skill level A glossary examples of calculations and equations and situational tasks can help you practice and understand chemistry This workbook also covers measurement chemical reactions and equations and matter elements compounds and mixtures Explore other aspects of the language including Formulas and ionic compounds Gases and the gas laws Atoms The mole elements and compounds Solutions and solution concentrations Chemical bonding Acids bases and buffers Practice makes perfect and whether you re taking lessons or teaching yourself CliffsStudySolver guides can help you make the grade **Workbook for Organic Chemistry** Jerry Jenkins, 2009-12-25 With authors who are both accomplished researchers and educators Vollhardt and Schore s Organic Chemistry is proven effective for making contemporary organic chemistry accessible introducing cutting edge research in a fresh student friendly way A wealth of unique study tools help students organize and understand the substantial information presented in this course And in the sixth edition the themes of understanding reactivity mechanisms and synthetic analysis to apply chemical concepts to realistic situations has been strengthened New applications of organic chemistry in the life sciences industrial practices green chemistry and environmental monitoring and clean up are incorporated This edition includes more than 100 new or substantially revised problems including new problems on synthesis and green chemistry and new challenging problems Chemistry John S. Phillips, Cheryl Wistrom, 2000 Problems and Solutions in Quantum Chemistry and Physics Charles Sidney Johnson, Lee G. Pedersen, 1986-01-01 Two hundred and eighty problems with detailed solutions plus 139 exercises all covering quantum mechanics wave mechanics angular momentum molecular spectroscopy scattering theory and related subjects An excellent problem book I would highly recommend it as a required supplement to students taking their first quantum chemistry course Journal of the American Chemical Society Modern Physics Wolfson, 1999 *Fundamentals of Physics, , Problem Supplement No. 1* Jearl Walker, 2001 This is a supplement to the text Fundamentals of Physics 6th Ed This supplement contains additional sample problems checkpoint style questions organizing questions discussion questions and new exercises and problems **Fundamentals of Physics, , Problem Supplement No. 1** David Halliday, Robert Resnick, Jearl Walker, 2000-06-06 No other book on the market today can match the success of Halliday Resnick and Walker s Fundamentals of Physics In a breezy easy to understand style the book offers a solid understanding of fundamental physics concepts and helps readers apply this conceptual understanding to quantitative problem solving Schaum's Outline of

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