

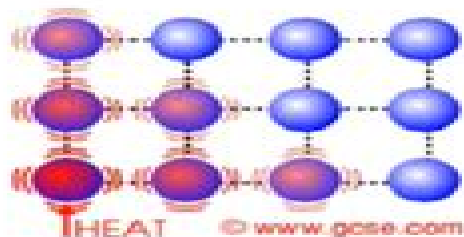
# CHAPTER 6

## TEMPERATURE & HEAT

### LESSON 1: TEMPERATURE & HEAT

#### Vibrating Matter

- All matter is made up of atoms that are moving, even solid objects which have compact molecular arrangement, have atoms that are continuously moving.



Heat is introduced one to atom of the molecule. Since these atoms are in lattice arrangement, the movement of vibration of one affects the others. Thus, this heat caused an increase in the kinetic energy of these atoms.

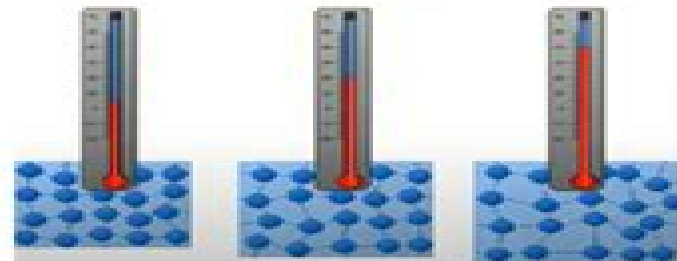
- The motion from the atoms gives the object energy.

#### Temperature

- The quantitative measure of the "hotness" or "coldness" of a system.

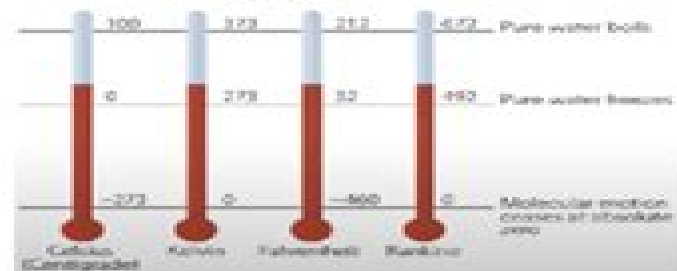
The introduction of heat to a particle, say an atom, results in an increase in the vibration or movement that is the increase in the kinetic energy of the system. This increase in kinetic energy gives rise to the increase in temperature.

- A measure of the average kinetic energy of the particles in an object.
- Clearly is dependent on particle movement.



#### Temperature Scales

- 4 basic temperature scales: namely (CFKR),
  - Celsius
  - Fahrenheit
  - Kelvin
  - Rankine



- **Kelvin** and **Rankine** scales are absolute temperature scales, meaning they don't have negative value readings.

# Chapter Lesson Temperature Change

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