

CHAPTER 7 REVIEW*Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

- 1.** C In a Stock name such as iron(III) sulfate, the roman numeral tells us _____.
 - (a) how many atoms of Fe are in one formula unit
 - (b) how many sulfate ions can be attached to the iron atom
 - (c) the charge on each Fe ion
 - (d) the total positive charge of the formula unit
- 2.** C The result of changing a subscript in a correctly written chemical formula is to _____.
 - (a) change the number of moles represented by the formula
 - (b) change the charges on the other ions in the compound
 - (c) change the formula so that it no longer represents the compound it previously represented
 - (d) have no effect on the formula
- 3.** The explosive TNT has the molecular formula $C_7H_5(NO_2)_3$.

<u>4 elements</u>	a. How many elements make up this compound?
<u>6 oxygen atoms</u>	b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$?
<u>21 atoms</u>	c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$?
<u>4.2×10^{24} atoms</u>	d. How many atoms are present in a sample of 2×10^{23} molecules of $C_7H_5(NO_2)_3$?
- 4.** How many atoms are present in each of these formula units?

<u>11 atoms</u>	a. $Ca(HCO_3)_2$
<u>45 atoms</u>	b. $C_{12}H_{22}O_{11}$
<u>10 atoms</u>	c. $Fe(ClO_2)_3$
<u>9 atoms</u>	d. $Fe(ClO_3)_2$
- 5.** N_2O_5 **a.** What is the formula for the compound dinitrogen pentoxide?

carbon(IV) sulfide **b.** What is the Stock name for the covalent compound CS_2 ?

 H_2SO_3 **c.** What is the formula for sulfurous acid?

phosphoric acid **d.** What is the name for the acid H_3PO_4 ?

Chapter 9 Chemical Names And Formulas Practice Problems Answers

Brendan G. Carr



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Barron's Chemistry Practice Plus: 400+ Online Questions and Quick Study Review Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2022-07-05 Barron's Chemistry Practice Plus features more than 400 online practice questions and a concise review guide that covers the basics of Chemistry. Inside you'll find concise review on the basics of Chemistry, an excellent resource for students who want a quick review of the most important topics, access to 400 online questions arranged by topic for customized practice, online practice includes answer explanations with expert advice for all questions plus scoring to track your progress. This essential guide is the perfect practice supplement for students and teachers.

Understanding Chemistry Fred M. Dewey, 1994 [General Organic and Biological Chemistry](#) Kenneth W. Raymond, 2009-12-14 This general organic and biochemistry text has been written for students preparing for careers in health-related fields such as nursing, dental hygiene, nutrition, medical technology, and occupational therapy. It is also suited for students majoring in other fields where it is important to have an understanding of the basics of chemistry. Students need have no previous background in chemistry but should possess basic math skills. The text features numerous helpful problems and learning features.

Student's Guide to Introduction to Chemical Principles by Edward I. Peters, 2d Ed Peter P. Berlow, 1978

Addison-Wesley Chemistry, 1990 **Study Guide to Accompany Basics for Chemistry** Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook. Each chapter contains Overview, Topical Outline, Skills, and Common Mistakes, which are all keyed to the textbook for easy cross reference. The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms, a summary of general concepts, and a list of numerical exercises, while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook. The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter, and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples. The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry, while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter. After briefly dealing with an overview of chemistry, this book goes on exploring the concept of matter, energy, measurement, problem solving, atom, periodic table, and chemical bonding. These topics are followed by discussions on writing names and formulas of compounds, chemical formulas, and the mole, chemical reactions, calculations based on equations, gases, and the properties of a liquid. The remaining chapters examine the solutions, acids, bases, salts, oxidation-reduction reactions, electrochemistry, chemical kinetics, and equilibrium, and nuclear, organic, and biological chemistry. This study guide will be of great value to chemistry teachers and students.

Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 *The Practice of Chemistry* Donald J. Wink, Sharon Fetzer-Gislason, Sheila

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