Nuclear Chemistry – Alpha, Beta, and Gamma Decay Practice

Identify the following examples as alpha, beta, or gamma decay.

1.
$$^{233}_{92}U \rightarrow ^{4}_{2}He + ^{229}_{90}Th _alpha_$$

4.
$$^{202}_{81}\text{TI} \rightarrow ^{202}_{81}\text{TI} + ^{0}_{0}\text{Y}$$
 gamma

2.
$$^{99}_{43}$$
Tc $\rightarrow ^{0}_{-1}$ e + $^{99}_{44}$ Ru beta

5.
$${}^{125}_{53}I \rightarrow {}^{0}_{0} Y + {}^{125}_{53}I$$
 gamma

3.
$$^{175}_{78}$$
 Pt $\rightarrow ^{171}_{76}$ Os + $^{4}_{2}$ He _alpha_

6.
$$^{82}_{35}Br \rightarrow ^{82}_{36}Kr + ^{0}_{-1}e$$
 beta

Alpha Decay: Write the alpha decay equations for the following nuclides. Example: ${}^{208}_{84}$ Po $\rightarrow {}^{4}_{2}$ He + ${}^{204}_{82}$ Pb

Example:
$${}^{208}_{84}$$
 Po $\rightarrow {}^{4}_{2}$ He + ${}^{204}_{82}$ Pb

7.
$$^{256}_{103}$$
Lr $\rightarrow ^{4}_{2}$ He + $^{252}_{101}$ Md

9.
$$^{185}_{79}$$
 Au $\rightarrow ^{4}_{2}$ He + $^{181}_{77}$ Ir

s.
$$^{231}_{91}$$
 Pa $\rightarrow ^{4}_{2}$ He + $^{227}_{89}$ Ac

10.
$$^{211}_{87}$$
 Fr $\rightarrow ^{4}_{2}$ He + $^{207}_{85}$ At

Beta Decay: Write the beta decay equations for the following nuclides.

Example:
$${}^{14}_{6}C \rightarrow {}^{14}_{7}N + {}^{0}_{-1}e$$

11.
$$^{24}_{11}$$
 Na $\rightarrow ^{24}_{8}$ O + $^{0}_{-1}$ e

13.
$$^{247}_{95}$$
 Am $\rightarrow ^{247}_{96}$ Cm + $^{0}_{-1}$ e

12.
$$^{201}_{79}$$
 Au $\rightarrow ^{201}_{80}$ Hg + $^{0}_{-1}$ e

14.
$${}^{90}_{38}$$
Sr $\rightarrow {}^{90}_{39}$ Y + ${}^{0}_{-1}$ e

Balancing Decay Reactions: Fill in the blank in each of the following decay reactions with the correct decay particle or decayed nucleus that will balance the reaction, also state whether it is alpha or beta decay

15.
$$^{184}_{74}$$
 W $\rightarrow ^{4}_{2}$ He + $^{180}_{72}$ Hf

20.
$$^{60}_{27}W \rightarrow ^{0}_{-1}e + ^{60}_{28}Ni$$

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Table of Contents Chemistry Nuclear Packet Worksheet Radioactivity Answers

- 1. Understanding the eBook Chemistry Nuclear Packet Worksheet Radioactivity Answers
 - The Rise of Digital Reading Chemistry Nuclear Packet Worksheet Radioactivity Answers
 - o Advantages of eBooks Over Traditional Books
- 2. Identifying Chemistry Nuclear Packet Worksheet Radioactivity Answers
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Nuclear Packet Worksheet Radioactivity Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chemistry Nuclear Packet Worksheet Radioactivity Answers
 - Personalized Recommendations
 - Chemistry Nuclear Packet Worksheet Radioactivity Answers User Reviews and Ratings
 - o Chemistry Nuclear Packet Worksheet Radioactivity Answers and Bestseller Lists
- 5. Accessing Chemistry Nuclear Packet Worksheet Radioactivity Answers Free and Paid eBooks
 - Chemistry Nuclear Packet Worksheet Radioactivity Answers Public Domain eBooks
 - Chemistry Nuclear Packet Worksheet Radioactivity Answers eBook Subscription Services
 - Chemistry Nuclear Packet Worksheet Radioactivity Answers Budget-Friendly Options

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 - o ePub, PDF, MOBI, and More
 - Chemistry Nuclear Packet Worksheet Radioactivity Answers Compatibility with Devices
 - Chemistry Nuclear Packet Worksheet Radioactivity Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Nuclear Packet Worksheet Radioactivity Answers
 - Highlighting and Note-Taking Chemistry Nuclear Packet Worksheet Radioactivity Answers
 - Interactive Elements Chemistry Nuclear Packet Worksheet Radioactivity Answers
- 8. Staying Engaged with Chemistry Nuclear Packet Worksheet Radioactivity Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Nuclear Packet Worksheet Radioactivity Answers
- 9. Balancing eBooks and Physical Books Chemistry Nuclear Packet Worksheet Radioactivity Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Nuclear Packet Worksheet Radioactivity Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Nuclear Packet Worksheet Radioactivity Answers
 - Setting Reading Goals Chemistry Nuclear Packet Worksheet Radioactivity Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Nuclear Packet Worksheet Radioactivity Answers
 - Fact-Checking eBook Content of Chemistry Nuclear Packet Worksheet Radioactivity Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements

• Interactive and Gamified eBooks

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