

Chapter 10 Review: Chemical Quantities & Calculations

(aka – mole & volume calculations, % composition, empirical & molecular formulas)

For the test, be sure that you are familiar, and able to complete, the problems below as well as the problems found on the following worksheets: "Molecular Weight and Mole Calculations," "Percent Comp. & Chemical Formula Calculations" and "Chapter 10 Sample Problems." Check the class website for answers to the review worksheet.

1. The air around you is made up of 20.95% oxygen. How many oxygen molecules are in 75.0 grams of air? The density of air is 1.19 g/L, *assume STP*.
2. How much helium (*grams*) would be needed to inflate a balloon to a volume of 5.50L? *assume STP*.
3. The molecular formula of an antibacterial drug is $C_{17}H_{15}FN_2O_5$. How many fluorine atoms are in 150mg of this drug?
4. A molecule is made up of 40.0% carbon, 6.67% hydrogen, and an unknown amount of oxygen. What is the molecular formula of the molecule if its molar mass is 90.0g/mole?
5. What is the empirical formula for a compound made up of 71.72% chlorine, 16.16% oxygen and 12.12% carbon? Calculate the molecular formula for the compound if it has a molar mass of 296.7g/mol.
6. How many molecules are in 8.42 Kilograms of C_2H_4 ?

Chemistry Packet Chapter 10 Chemical Quantities

Practice Problems Answers

Steven S. Zumdahl



Chemistry Packet Chapter 10 Chemical Quantities Practice Problems Answers:

Basic Chemistry Steven S Zumdahl,1996 *Addison-Wesley Chemistry* ,1990 *Introductory Chemistry* Steven S. Zumdahl,1996 *Study guide to accompany chemistry* Karen Timberlakes,1983 *Study Guide to Accompany Basics for Chemistry* Martha Mackin,2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students *Chemistry* John S. Phillips,Cheryl Wistrom,2000 *The Practice of Chemistry* Donald J. Wink,Sharon Fetzer-Gislason,Sheila McNicholas,2003-03 Students can't do chemistry if they can't do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book's unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt 832KB **Fundamentals of Chemistry** Ralph A. Burns,2003 For one semester preparatory chemistry courses or general purpose introductory chemistry courses This clearly written well illustrated versatile textbook provides thorough coverage of chemistry with a balance of problem solving skills real world applications and an emphasis on critical thinking and the process of science A supporting theme throughout the text continually emphasizes that chemistry is everywhere **Prentice Hall Chemistry** ,2000 **Schaum's Outline of Theory and Problems of Chemistry Foundations** David E. Goldberg,1991 A solved problem Outline for students with no

high school chemistry background or a poor course in high school chemistry All topics are presented at an elementary level to commensurate with the introductory course simpler math is assumed developed a less intense approach to the basic concepts of chemistry a simpler manner of presentation There are hundreds of solved supplementary problems Journal of Chemical Education ,1924 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers CHEMISTRY SILBERBERG,2003 **Essentials of Chemistry** Ralph A. Burns,John William Hill,1995 The second edition of this chemistry textbook that uses practice examples and applications relating chemistry to our lives and the environment **Holt Chemistry** Ralph Thomas Myers,2004 Chemistry and Toxicology of Pollution Des W. Connell,Gregory J. Miller,2022-11-29 Describes the transport of pollutants through the environment and their impact on natural and human systems fully updated to cover key topics in modern pollution science Chemistry and Toxicology of Pollution examines the interactions and adverse effects of pollution on both natural ecosystems and human health addressing chemical toxicological and ecological factors at both the regional and global scale The book is written using a conceptual framework that follows the interaction of a pollutant with the environment from distribution in the various abiotic sectors of the environment to exposure and effects on individuals and ecosystems The authors also highlight the critical role of various socio economic political and cultural aspects in achieving sustainable goals strategies and science based solutions to pollution and health This comprehensive volume covers the chemical behavior and governing principles of pollutants their interactions with humans and ecosystems and the methods and processes of environmental risk assessment and pollution management Extensively revised and expanded the second edition equips readers with the knowledge required to help lead the way towards a healthy and sustainable future New chapters address current pollution issues such as global warming and climate change recent advances in environmental science the monitoring and evaluation of new and emerging pollutants risk assessment and remediation and innovative pollution management approaches and techniques With in depth material on human toxicology integrated throughout the text Chemistry and Toxicology of Pollution Provides an effective framework for interpreting the information produced by international national and local agencies Presents unifying theories and principles supported by up to date scientific literature Offers broad coverage of pollution science with an emphasis on North America the UK Europe China India and Australia Discusses the similarities and differences of the impact of pollutants on the natural environment and humans Chemistry and Toxicology of Pollution Second Edition enables readers to view pollution in its correct perspective and develop appropriate control measures It is essential reading for scientists academic researchers policymakers professionals working in industry and advanced students in need of a clear understanding of the nature and effects of environmental pollution **Journal of the Society of Chemical Industry** ,1924 **The Chemical World** John C. Kotz,1994 **Understanding Chemistry** Fred M. Dewey,1994 The Publisher ,1910 **The Engineer** ,1877

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