

1) List the most common organic functional groups that function as acids or bases. Write an equation of each with HCl or NaOH to show their behavior as acids or bases. (You will also enter these functional groups in the table in Question 2.)



2) Fill in the following table showing the general solubility behavior of the organic acids and bases you listed above. Assume that they are not small enough to have significant water solubility.

Type of Organic Compound	Functional Group	Soluble in Water?	Soluble in 2 M NaHCO ₃ ?	Soluble in 2 M NaOH?	Soluble in 2 M HCl?
Neutral	Many				
Acidic	Carboxylic acids	Yes	Yes	Yes	No
Weakly Acidic	Phenols	Yes	Yes	Yes	Yes
Basic	Amines	No	Yes	Yes	Yes

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