



# IONIC AND METALLIC BONDING

## Chapter Test A

### A. Matching

Match each description with the correct term or formula. Write the term or formula in the space provided.

Column A	Column B
_____ 1. Anions are called _____.	_____ a. positive electron and positive electron and a positive electron
_____ 2. Anions are called _____.	_____ b. the electron configuration about valence shell and electron
_____ 3. Anions are called _____.	_____ c. the electron configuration about valence shell and electron
_____ 4. Anions are called _____.	_____ d. the electron configuration about valence shell and electron
_____ 5. Anions are called _____.	_____ e. a diagram of valence electron about the nucleus of an atom
_____ 6. Anions are called _____.	_____ f. group and it is called valence
_____ 7. Anions are called _____.	_____ g. the most electronegative
_____ 8. Anions are called _____.	_____ h. the electronegativity is a good way to find out how
_____ 9. Anions are called _____.	_____ i. the electronegativity is a good way to find out how
_____ 10. Anions are called _____.	_____ j. the electronegativity is a good way to find out how
_____ 11. Anions are called _____.	_____ k. the electronegativity is a good way to find out how
_____ 12. Anions are called _____.	_____ l. the electronegativity is a good way to find out how

### B. Multiple Choice

1. Which of the following is not a correct statement?

- \_\_\_\_\_ a. The electronegativity of an atom is a measure of its ability to attract electrons.  
 \_\_\_\_\_ b. The electronegativity of an atom is a measure of its ability to attract electrons.  
 \_\_\_\_\_ c. The electronegativity of an atom is a measure of its ability to attract electrons.  
 \_\_\_\_\_ d. The electronegativity of an atom is a measure of its ability to attract electrons.

# Chapter 7 Ionic And Metallic Bonding Practice Test

**Kathy A. Zahler**



## **Chapter 7 Ionic And Metallic Bonding Practice Test:**

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This book delves into Chapter 7 Ionic And Metallic Bonding Practice Test. Chapter 7 Ionic And Metallic Bonding Practice Test is a crucial topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Chapter 7 Ionic And Metallic Bonding Practice Test, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Chapter 7 Ionic And Metallic Bonding Practice Test
- Chapter 2: Essential Elements of Chapter 7 Ionic And Metallic Bonding Practice Test
- Chapter 3: Chapter 7 Ionic And Metallic Bonding Practice Test in Everyday Life
- Chapter 4: Chapter 7 Ionic And Metallic Bonding Practice Test in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Chapter 7 Ionic And Metallic Bonding Practice Test. The first chapter will explore what Chapter 7 Ionic And Metallic Bonding Practice Test is, why Chapter 7 Ionic And Metallic Bonding Practice Test is vital, and how to effectively learn about Chapter 7 Ionic And Metallic Bonding Practice Test.

3. In chapter 2, the author will delve into the foundational concepts of Chapter 7 Ionic And Metallic Bonding Practice Test. This chapter will elucidate the essential principles that must be understood to grasp Chapter 7 Ionic And Metallic Bonding Practice Test in its entirety.

4. In chapter 3, this book will examine the practical applications of Chapter 7 Ionic And Metallic Bonding Practice Test in daily life. The third chapter will showcase real-world examples of how Chapter 7 Ionic And Metallic Bonding Practice Test can be effectively utilized in everyday scenarios.

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The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Chapter 7 Ionic And Metallic Bonding Practice Test.

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### Chapter 7 Ionic And Metallic Bonding Practice Test Introduction

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example 1 if  $\overline{BD}$  is an angle bisector find  $\angle ADB$   $\angle BDC$  since the angle bisector cuts the angle in half the other half must also measure 55  $\angle ADB = 55^\circ$   $\angle BDC = 55^\circ$

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point  $d$  is on side  $bc$  side  $ac$  is five point nine units side  $db$  is two point eight units side  $ab$  is eight point one units

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and  $abstvxyz$   $cd$  bisects  $ab$  4 draw acute angle  $e$  construct the bisector of  $e$   $tm$  bisects  $5x4$  and  $6x2$   $x^2 - 5$  algebra  $stu$  so that  
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