

Oxoacids (O—H bond): contain H, O, and a central non-metal
 Acidity determined by the polarity of the O—H bond:

1. central atom from the same group of the periodic table, same oxidation number

Relative strength increases with increased electronegativity of the central atom.



2. same central atom, different oxidation number (# of O's)

Higher oxidation number = stronger acid.



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