

Acids & Bases: Assessment Review

Name _____

Goal – The student will be able to identify the acid, base, conjugate acid, and conjugate base in an acid – base reaction.

1. Write a chemical equation showing how HCO_3^- behaves as an acid when dissolved in water. Label the acid, base, conjugate acid & conjugate base.

2. Write a chemical equation showing how ammonia (NH_3) behaves as a base when dissolved in water. Label the acid, base, conjugate acid & conjugate base.

Goal – The student will be able to calculate with the $[\text{H}^+]$ or $[\text{OH}^-]$ from the ion product (K_w) or pH from pOH (a one-step pH block calculation).

3. Recreate the calculation flowchart on the diagram provided. (Be careful...it's been flipped!)



4. What is $[\text{OH}^-]$ if a solution has $[\text{H}^+] = 7.1 \times 10^{-4} \text{ M}$? Is the solution acidic, basic, or neutral?

5. Calculate $[\text{H}^+]$ in a solution if $[\text{OH}^-] = 5.1 \times 10^{-11} \text{ M}$. Is the solution acidic, basic, or neutral?

6. Calculate the $[\text{H}^+]$ in a solution that shows a pH of 11.75. Is the solution acidic, basic, or neutral?

7. What is the pOH of a solution if $[\text{OH}^-] = 5.0 \times 10^{-4} \text{ M}$? Is the solution acidic, basic, or neutral?

Chapter 8 Solutions Acids Bases Assessment Answer Key

Peter P. Berlow



Chapter 8 Solutions Acids Bases Assessment Answer Key:

Student Study Guide Fundamentals of Chemistry by David E. Goldberg Margaret G. Kimble, 1994 **Polymer Concretes** Mostafa Hassani Niaki, Morteza Ghorbanzadeh Ahangari, 2022-12-30 Polymer Concretes Advanced Construction Materials provides a comprehensive study on polymer concrete PC discussing historical perspectives of its use the classification and applications of PC and the advantages and disadvantages of its use Materials such as resin aggregates micro fillers fibers and nanofillers are systematically summarized as well as their effects on PC Also examined are the properties fabrication methods and the standards for testing the material properties as well as the future outlook for PC applications This book Investigates the various properties of PC Covers the physical mechanical thermal chemical electrical and environmental properties of PC Examines fabrication methods standards for testing and the future outlook for various applications The book is ideal for students taking related courses in Civil Mechanical Chemical and Material Engineering It also serves as a useful guide for researchers in the areas of concrete and construction materials composites and nanocomposites and advanced materials as well as professionals working in fields such as construction precast concrete products manufacture transportation and road construction architecture and more **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science**, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Visualizing Everyday Chemistry** Douglas P. Heller, Carl H. Snyder, 2015-01-20 Visualizing Everyday Chemistry Binder Ready Version is for a one semester course dedicated to introducing chemistry to non science students It shows what chemistry is and what it does by integrating words with powerful and compelling visuals and learning aids With this approach students not only learn the basic principles of chemistry but see how chemistry impacts their lives and society The goal of Visualizing Everyday Chemistry Binder Ready Version is to show students that chemistry is important and relevant not because we say it is but because they see it is This text is an unbound binder ready version

Experiments for Living Chemistry David Ucko, 2012-12-02 Experiments for Living Chemistry provides practical hands on experiments illustrating the concepts substances and techniques that are important to students in the health related sciences Many of these experiments are based on physiological substances to show students how chemical principles apply to the functioning of their own bodies while other experiments use cut outs to help students visualize such complex concepts as bonding and protein synthesis This book is organized into 23 chapters that correspond on a chapter by chapter basis with the Living Chemistry textbook The first five chapters include discussions on matter measurement chemical bonding compounds chemical change gases and respiration The subsequent chapters deal with water solutions acids bases salts hydrocarbons

and nuclear and organic chemistry Other chapters explore the oxygen and other derivatives of the hydrocarbons carbohydrates lipids proteins enzymes and digestion Considerable chapters are devoted to the metabolism of carbohydrate energy lipid and proteins The remaining chapters examine the heredity and protein synthesis vitamins hormones body fluids drugs and poisons At the end of each chapter there are sets of questions designed to help the student relate the laboratory experiments to the textbook and to the lecture portion of the course Each experiment in the chapter has a corresponding question set that should be answered only after the experiment has been completed This book is an invaluable study guide to chemistry teachers and undergraduate students Instructor's Manual and Test Bank to Accompany Basic Concepts of

Chemistry Malone,2000-04-18 *A Laboratory Manual of General Chemistry* James Henri Walton,Francis Craig Krauskopf,1920 *Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today* Seager/Slabaugh,2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application *Study Guide for Atoms, Molecules, and Life* Michael S. Matta,1981 *Student's Guide to Introduction to Chemical Principles by Edward I. Peters, 2d Ed* Peter P. Berlow,1978 *Industrial & Engineering Chemistry*,1923 *Chemical Interactions*,2005 Teaching High School Science Alexander Joseph,1961 Fundamentals of Chemistry Lotta Jean Bogert,1924 **Chemical Engineering Catalog**,1920 **Study Guide for Chemical Principles, Fourth Edition, by Dickerson, Gray, Darensbourg, and Darensbourg** Patricia L. Samuel,1984 **Industrial and Engineering Chemistry**,1923 **A Text-book of chemistry and chemical uranalysis for nurses** Harold Lindsay Amoss,1915 **Introductory Chemistry for Today** Spencer L. Seager,Michael R. Slabaugh,2004 Distinguished by its superior allied health focus and integration of technology Seager and Slabaugh s INTRODUCTORY CHEMISTRY FOR TODAY Fifth Edition continues to lead the market on both fronts through numerous allied health related applications examples boxes and a new Companion Web Site GOB ChemistryNow tm In addition to the many resources found in GOB ChemistryNow this powerful new Web site contains questions modeled after the Nursing School and Allied Health Entrance Exams and NCLEX LPN Certification Exams The authors strive to dispel users inherent fear of chemistry and to instill an appreciation for the role chemistry plays in our daily lives through a rich pedagogical structure and an accessible writing style that provides lucid explanations In addition Seager and Slabaugh s CHEMISTRY FOR TODAY Fifth Edition provides greater support in both problem solving and critical thinking skills By demonstrating how this information will be important to a reader s future career and providing important career information online the authors not only help readers to set goals but also to focus on achieving them **Exercises in Second Year Chemistry** William Henry Chapin,1944

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