

Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
Li	1 ve	transfers 1ve <sup>-</sup> to F	Li <sup>+</sup>
F	7 ve	accepts 1ve <sup>-</sup> from Li	F <sup>-</sup>

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
1) $\text{Li} + \text{Cl} \Rightarrow \text{LiCl}$ 	Li	1 ve	donate 1e <sup>-</sup>	Li <sup>+</sup>
	Cl	1 ve	accepts 1e <sup>-</sup>	Cl <sup>-</sup>
2) $\text{Ca} + \text{O} \Rightarrow \text{CaO}$ 	Ca	2 ve	donates 2e <sup>-</sup>	Ca <sup>+2</sup>
	O	6 ve	accepts 2e <sup>-</sup>	O <sup>-2</sup>
3) $\text{Be} + \text{F} \Rightarrow \text{BeF}_2$ 	Be	2 ve	donates 2e <sup>-</sup> to each F	Be <sup>+2</sup>
	F	7 ve	each F accepts 1ve each	F <sup>-</sup>
4) $\text{Mg} + \text{S} \Rightarrow \text{MgS}$ 	Mg	2 ve	donates 2e <sup>-</sup> to S	Mg <sup>+2</sup>
	S	6 ve	accepts 2e <sup>-</sup>	S <sup>-2</sup>
5) $\text{K} + \text{F} \Rightarrow \text{KF}$ 	K	1 ve	donates 1ve <sup>-</sup>	K <sup>+1</sup>
	F	7 ve	accepts 1ve <sup>-</sup>	F <sup>-1</sup>

# Chapter Ionic Metallic Bonding Worksheet Answers

**Robert W. Avakian**



## **Chapter Ionic Metallic Bonding Worksheet Answers:**

Chemical Misconceptions Keith Taber, 2002 Part 2 provides strategies for dealing with some of the misconceptions that students have by including ready to use classroom resources     **Addison-Wesley Science Insights** ,1996     Exploring Earth and Space Michael DiSpezio, 1995 A textbook exploring such aspects of matter and energy as heat electricity and nuclear chemistry with suggested activities and review questions at the end of each chapter     *Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science* ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction     **Science Interactions** Robert W. Avakian, 1995-07-17     **Teaching Chemical Bonding** Margaret Irene Lindsay, 1995 This document presents an instructional strategy for teaching chemical bonding using parables and music Games student interactions and worksheets are included in the lesson plans Topics include metallic bonding covalent bonding including molecular and network structure and ionic bonding JRH     **Chemical Bonding** Kok Koon Leong, 2021-05-03 This book introduces the principles behind chemical bonding to teenagers between the ages of fifteen to seventeen Topics covered include ionic bonding covalent bonding and metallic bonding     **Atoms & Chemical Bonding Science Learning Guide** NewPath Learning, 2014-03-01 The Atoms Atomic Configuration Chemical Bonding Ionic Bonding Ionic Compounds Covalent Bonding Covalent Compounds Naming Compounds and Metallic Bonding Aligned to Next Generation Science Standards NGSS and other state standards

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