

## Applying Scientific Methods

A college chemistry student is studying the properties of four unknown compounds, W, X, Y, and Z. She has been informed that one of them is ionic and that the other three are covalent. Of the latter, she has been told that the attractions between the formula units are dispersion forces in one case, hydrogen bonds in another, and covalent (network) bonds in another. She has been asked to determine the type of attractions for each substance. In an attempt to do so, she carries out experiments that reveal information about the substances' properties. Use this information to answer the questions that follow.

| Property                             | Unknown W     | Unknown X     | Unknown Y     | Unknown Z     |
|--------------------------------------|---------------|---------------|---------------|---------------|
| Hardness of solid                    | soft          | soft          | brittle       | brittle       |
| Melting point ( $^{\circ}\text{C}$ ) | -10           | -200          | 1500          | 3000          |
| Boiling point ( $^{\circ}\text{C}$ ) | 120           | -150          | 2500          | 4500          |
| Solubility in polar solvent          | very soluble  | insoluble     | very soluble  | insoluble     |
| Solubility in nonpolar solvent       | insoluble     | very soluble  | insoluble     | insoluble     |
| Conductivity of solid                | nonconducting | nonconducting | nonconducting | nonconducting |
| Conductivity of liquid               | nonconducting | nonconducting | conducting    | nonconducting |

- On the basis of the results of her hardness test, the student concludes that unknowns W and X are covalent substances, and that either Y or Z is the ionic substance. Comment on the soundness of her conclusion, given only the hardness information.

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# Chemistry Chapter Assessment Applying Scientific Methods Answers

**J Spring**



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