

4. a. A sample of a  $\text{CaCl}_2 \cdot 2\text{H}_2\text{O} / \text{K}_2\text{C}_2\text{O}_4 \cdot \text{H}_2\text{O}$  solid salt mixture is dissolved in ~150 mL of deionized water, adjusted to a pH that is basic. The precipitate, after having been filtered, was air-dried and weighed. Data for Trials 1 and 2 were obtained as shown. Complete the following table. (See *Report Sheet*.) Record calculated values with the correct number of significant figures.

**Calculation Zone**

**A. Precipitation of  $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$  from the Salt Mixture**

- Mass of salt mixture (g)
- Mass of filter paper (g)
- Mass of filter paper and  $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$  (g)
- Mass of air-dried  $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$  (g)

0.879  
1.896  
2.180  
0.239

**Data Analysis, 1.**

**B. Determination of Limiting Reactant**

- Limiting reactant in salt mixture
- Excess reactant in salt mixture

$\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$   
 $\text{K}_2\text{C}_2\text{O}_4 \cdot \text{H}_2\text{O}$

**Data Analysis, 3.**

**Data Analysis**

- Moles of  $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$  precipitated (mol)  
Show calculation.
- Moles of limiting reactant in salt mixture (mol)  
See equation 8.1.
- Mass of limiting reactant in salt mixture (g)  
Show calculation.
- Mass of excess reactant in salt mixture (g)  
Equals mass of salt mixture minus mass of limiting reactant.  
Show calculation.
- Percent limiting reactant in salt mixture (%)  
Show calculation.
- Percent excess reactant in salt mixture (%)

0.0012  
0.0012  
0.12  
0.12  
0.12  
0.12

**Data Analysis, 4.**

**Data Analysis, 5.**

4. b. For Trials 2 and 3, the percent  $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$  in the mixture was 39.7% and 28.4% respectively.

a. What is the average percent of  $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$  in the sample? **Data Analysis, B.**

b. What are the standard deviation and the relative standard deviation (%RSD) for the percent of  $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$  in the sample? **Data Analysis, C and D.**

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**Jo Allan Beran**



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