

Hints & Solutions



Physical World, Units and Measurements



Topic-1: Unit of Physical Quantities



(a) Density of material in SI unit,

$$= \frac{128 \text{kg}}{\text{m}^3}$$
Density of material in new system
$$= \frac{128(50 \text{ g})(20)}{(25 \text{ cm})^3 (4)^3} = \frac{128}{64}(20) = 40 \text{ units}$$

(b) According to question 2.

 $E_{\nu} \propto J_{\nu} B_{\nu}$

... Constant of proportionality
$$K = \frac{E_y}{B_Z J_x} = \frac{C}{J_x} = \frac{m^3}{As}$$

- [As $\frac{E}{R} = C$ (speed of light) and $J = \frac{1}{Area}$]
- (a, b, c, d)

(a) Inductance,
$$L = \frac{Flux(\phi)}{Current(I)} = \frac{weber}{ampere} \Rightarrow benry$$

(b)
$$\epsilon = L\left(\frac{dt}{dt}\right) \Rightarrow [L] = \frac{\epsilon}{dt/dt} = \frac{\text{volt} \times \text{sec}}{\text{smp}} \Rightarrow H = \frac{Vs}{A}$$

(c)
$$U = \frac{1}{2}LI^2 \Rightarrow L = \frac{2U}{I^2} = \frac{J}{A^2} \Rightarrow H = \frac{J}{A^2}$$

(d)
$$L = \frac{\epsilon}{dt/dt} = \left(\frac{\epsilon}{dt}\right)dt = \text{ohm} \times \text{sec}$$

 $\therefore \text{ henry} = \text{ohm} - \text{sec}$
 $A \rightarrow p, q$

Using
$$F = \frac{GM_gM_g}{g^2} \Rightarrow GM_gM_g = Fr^2 = Nm^2 = kg\frac{m}{s^2} \times m^2$$

= $kg\frac{m}{s^2} \times m^2$

Also (volt) (coulomb) (metre) = (joule) (metre) = $kg \text{ m}^3 s^{-2}$ = (N-m) (m) = $Nm^2 = kg \text{ m}^3 s^{-2}$ B $\rightarrow r_* s$

Using
$$v_{max} = \sqrt{\frac{3RT}{M}} \implies v_{max}^2 = \frac{3RT}{M}$$

Unit of $\frac{3RT}{M}$ is $m^2 s^{-2}$

Also (farad) (volt)² (kg)⁻¹ = (joule) kg⁻¹ ($\because u = \frac{1}{2}cv^2$) = N-m kg⁻¹ = kg ms⁻² m kg⁻¹ = m²s⁻²

$$C \rightarrow r, s$$

Using
$$F = qvB \implies v^2 = \frac{F^2}{-2n^2}$$

Using $F = qvB \implies v^2 = \frac{F^2}{q^2B^2}$ \therefore Unit of v^2 is m^2s^{-2} which is further equal to FF^2 kg⁻¹. $\mathbf{D} \rightarrow \mathbf{r}$, s

Reason : Escape velocity $v_g = \sqrt{\frac{2GM}{B}} \implies v_g^2 = \frac{2GM}{B}$

- ... Unit of GM is m² s⁻².
- Capacitance $C = \frac{q}{v} = \frac{q}{to}$ coulomb-volt 1, coulomb2-joule 1

Inductance $\frac{L}{R} = t$ and $R = \frac{v}{t}$ ohm-sec, volt-second (ampere)⁻¹

Magnetic Induction $F = IIB \Rightarrow B = \frac{F}{rs}$ newton (ampere-metre)⁻¹

- (i) The M.K.S. unit of Young's modulus $\left(Y = \frac{F}{4} / \frac{\Delta I}{L}\right)$ is
 - (ii) The M.K.S. unit of magnetic induction $B = \frac{6}{4}$ is tesla
 - (iii) The M.K.S. unit of power of lens $P = \frac{1}{f(\text{in metre})}$ is dioptre or m-1.



Topic-2: Dimensions of Physical Quantities



(a) We know that

Speed of light,
$$c = \frac{1}{\sqrt{\mu_0 \epsilon_0}} = x$$

Also, $c = \frac{E}{R} = y$

Time constant, $\tau = Rc = t$

$$\therefore z = \frac{l}{Rc} = \frac{l}{t} = \text{Speed}$$

Thus, x, y, z will have the same dimension of speed.

2.

(d) From Soemula,
$$\frac{dQ}{dt} = kA \frac{dT}{dx}$$

$$\Rightarrow k = \frac{\left(\frac{dQ}{dt}\right)}{A\left(\frac{dT}{dx}\right)}, \{k\} = \frac{[ML^2T^{-3}]}{[L^2][KL^{-1}]} = [MLT^{-3}K^{-1}]$$

(c) Dimension of Force F = M¹L¹T⁻²

Dimension of velocity $V = L^{\dagger}T^{-\dagger}$

Dimension of work $= M^{0}L^{2}T^{-2}$

Dimension of length = L.

Moment of inertia = ML2

$$\therefore x = \frac{B^2 V^2}{B^2 Z^4} = \frac{(M^2 L^2)(M^2 L^3 T^{-2})(L^2 T^{-2})^2}{(M^2 L^2 T^{-2})(L^4)}$$

$$= \frac{M^2 L^{-2} T^{-2}}{L^2} = M^4 L^{-3} T^{-2} = \text{Energy density}$$

(b) Solar constant = Energy Time Area

Dimension of Energy, $E = ML^2T^{-2}$

Dimension of Time - T Dimension of Area - L²

... Dimension of Solar constant

$$= \frac{M^1L^2T^{-2}}{TL^2} = M^1L^0T^{-3}.$$

(d) Young's modulus, $Y = \frac{\text{stress}}{\text{strain}}$

$$\Rightarrow Y = \frac{F}{A} / \frac{\Delta \ell}{\ell_0} = FA^{-1}V^0$$

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P. VENUGOPALA RAO

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Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction Prentice Hall Grammar Jeanette Adkins, 2003 Hall Math Course 3 Adapted Student Workbook 2007c Prentice-Hall Staff, 2006-07-15 Prentice Hall Mathematics Course 3 A solid foundation preparing students for Algebra 1 Chapters 1 3 provide a solid foundation of integers rational numbers and real numbers setting the stage for equations inequalities and functions Real World applications to the more abstract algebraic concepts are found throughout the text An average of over five Activity Labs per chapter ensures students receive the visual and special instruction necessary to conceptualize these abstract concepts better preparing them for advanced math courses Prentice Hall Science Series, 1994 Prentice-Hall Staff, 1993 The Accounting Review William Andrew Paton, 1963 Includes section Reviews **Resources for Teaching Middle School Science** Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and

multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources. One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences. Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance. Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents.

Prentice Hall Chemistry, 2000

TEXTBOOK OF ENVIRONMENTAL ENGINEERING P. VENUGOPALA RAO,2002-01-01 Designed for a first course in environmental engineering for undergraduate engineering and postgraduate science students the book deals with environmental pollution and its control methodologies It explains the basic environmental technology environmental sanitation water supply waste management air pollution control and other related issues and presents a logical and systematic treatment of topics The book an outgrowth of author s long experience in teaching the postgraduate science and engineering students is presented in a student oriented approach It is interspersed with solved examples and illustrations to reinforce many of the concepts discussed and apprise the readers of the current practices in areas of water processing water distribution collection and treatment of domestic sewage and industrial waste water and control of air pollution It emphasizes fundamental concepts and basic appli cations of environmental technology for management of environmental problems Besides students the book will be useful to the academia of environmental sciences civil environmental engineering as well as to environmentalists and administrators working in the field of pollution control

Books in Print, 1959**

Handbook for Teaching Introductory Psychology Michelle Rae Hebl, Charles L. Brewer, Ludy T. Benjamin, Jr., 2001-08-01 Like its predecessors Volume III of the Handbook for Teaching Introductory Psychology provides introductory psychology instructors with teaching ideas and activities that can immediately be put into practice in the classroom It contains an organized collection of articles from Teaching of Psychology TOP the official journal of the Society for the Teaching of Psychology Division 2 of the American Psychological Association Volume III contains 89 articles from TOP that have not been included in other volumes Another distinction between this volume and its predecessors is its emphasis on testing and assessment The book is divided into two sections Section One Issues and Approaches in Teaching Introductory Psychology contains 52 articles on critical issues such as how to approach the course understanding students interests perceptions and motives students existing knowledge of psychology including their misconceptions a comparison of introductory textbooks and tips on how to evaluate them test questions and student factors affecting exam performance an overview of different forms of feedback giving extra credit and how to deal with academic dishonesty Section Two consists of 37 articles that present demonstrations class and laboratory projects and other techniques to enhance teaching and learning

in both the introductory as well as advanced courses in the discipline This section is organized so as to parallel the order of topics found in most introductory psychology textbooks Intended for academicians who teach the introductory psychology course and or oversee grad assistants who teach the course all royalties of the book go directly to the Society for the Teaching of Psychology to promote its activities to further improve the teaching of psychology

Student Guide and

Review Manual John K. Harris, 1999-07

Vocational-technical Learning Materials Bruce Reinhart, 1974

Student-involved Classroom Assessment Richard J. Stiggins, 2001 For courses in Classroom Assessment Test and Measurement and training of teachers This core text for courses in classroom assessment is the leading text for courses where the focus is squarely on teaching teachers and preservice teachers how to assess students in their day to day classrooms It takes a stronger focus than any other book on integrating assessment and instruction And it provides very clear guidance on how to construct all types of items and tests but also explains better than any other text what each type of item can and cannot assess Writing and Grammar ,2001 Expressways Student Course Steven J. Molinsky, Bill Bliss, 1996-07 A beginning course for students of English which integrates lifeskill topics functions and grammar in an imaginative highway The Catholic School Journal ,1967 Fundamentals of General, Organic, and Biological Chemistry Susan theme McMurry, 2006-07 The Study Guide and Full Solutions Manual explain in detail how the answers in text and end of chapter problems are obtained They also contain chapter summaries study hints and self tests for each chapter MOUS Test Preparation Guide for Excel 2000 Emily Ketcham, Carolyn Monroe, 2001 Based on the actual MOUS exams the streamlined PH MOUS Test Prep Guide text software packages coach students through successful completion of the MOUS certification exams from Core to Expert skill levels Dynamically interactive enhanced by the unparalleled MOUS PinPoint 2000 software the series takes students step by step to guaranteed success Each chapter focuses on specific learning objectives based on the Core and Expert MOUS objectives discusses specific skills necessary to complete the MOUS certification exam and walks students through condensed step by step mini tutorials for a specific skill set Students then use the interactive program on the student MOUS PinPoint interactive CD ROM included in every copy of the text to see the material demonstrated to complete in text projects and to practice for the MOUS exams Catalog of Copyright Entries. Third Series Library of Congress. Copyright Office, 1952 Includes Part 1A Books and Part 1B Pamphlets Serials and Contributions to Periodicals The Accountant's Magazine ,1965

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