STOICHIOMETRY

SECTION 12.1 THE ARITHMETIC OF EQUATIONS (pages 353-358)

This section explains how to calculate the amount of reactants required or product formed in a nonchemical process. It teaches you how to interpret chemical equations in terms of interacting moles, representative particles, masses, and gas volume at STP.

Using Everyday Equations (pages 353-355)

1. How can you determine the quantities of reactants and products in a chemical reaction?

You can use the balanced equation.

- Quantity usually means the _______ of a substance expressed in grams or moles.
- A bookcase is to be built from 3 shelves (Sh), 2 side boards (Sb), 1 top (T), 1 base (B), and 4 legs (L). Write a "balanced equation" for the construction of this bookcase.

3Sh + 2Sb + T + B + 4L = Sh.Sb.TBL.

Using Balanced Chemical Equations (page 354)

- Is the following sentence true or false? Stoichiometry is the calculation of quantities in chemical reactions.
- 5. Calculations using balanced equations are called __stoichiometric calculations

► Interpreting Chemical Equations (pages 356-357)

6. From what elements is ammonia produced? How is it used?

Ammonia molecules are composed of nitrogen and hydrogen; it is used as a fertilizer.

Circle the letter of the term that tells what kind of information you CANNOT get from a chemical equation.

a. moles

d. volume

b. mass

e. number of particles

c) size of particles

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Charles H. Corwin

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