

Name: _____

Balancing Chemical Equations

Directions: Balance each of the following chemical equations.



- 1) ___ H_3PO_4 + ___ KOH \rightarrow ___ K_3PO_4 + ___ H_2O
- 2) ___ K + ___ B_2O_3 \rightarrow ___ K_2O + ___ B
- 3) ___ HCl + ___ NaOH \rightarrow ___ NaCl + ___ H_2O
- 4) ___ Na + ___ NaNO_3 \rightarrow ___ Na_2O + ___ N_2
- 5) ___ C + ___ S_8 \rightarrow ___ CS_2
- 6) ___ Na + ___ O_2 \rightarrow ___ Na_2O
- 7) ___ N_2 + ___ O_2 \rightarrow ___ N_2O_5
- 8) ___ H_3PO_4 + ___ $\text{Mg}(\text{OH})_2$ \rightarrow ___ $\text{Mg}_3(\text{PO}_4)_2$ + ___ H_2O
- 9) ___ NaOH + ___ H_2CO_3 \rightarrow ___ Na_2CO_3 + ___ H_2O
- 10) ___ KOH + ___ HBr \rightarrow ___ KBr + ___ H_2O
- 11) ___ Na + ___ O_2 \rightarrow ___ Na_2O
- 12) ___ $\text{Al}(\text{OH})_3$ + ___ H_2CO_3 \rightarrow ___ $\text{Al}_2(\text{CO}_3)_3$ + ___ H_2O
- 13) ___ Al + ___ S_8 \rightarrow ___ Al_2S_3
- 14) ___ Cs + ___ N_2 \rightarrow ___ Cs_3N
- 15) ___ Mg + ___ Cl_2 \rightarrow ___ MgCl_2
- 16) ___ Rb + ___ RbNO_3 \rightarrow ___ Rb_2O + ___ N_2
- 17) ___ C_6H_6 + ___ O_2 \rightarrow ___ CO_2 + ___ H_2O
- 18) ___ N_2 + ___ H_2 \rightarrow ___ NH_3

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**American Association of Petroleum
Geologists**



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Teacher Instruction and Student Performance in Balancing Chemical Equations William Levi Yarroch,1983

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