

Name: \_\_\_\_\_ Date: \_\_\_\_\_ Period: \_\_\_\_\_

## Balancing Chemical Equations Using Models

### Objective

Use models as a means for balancing chemical equations.

### Materials

For each group of 4 students:

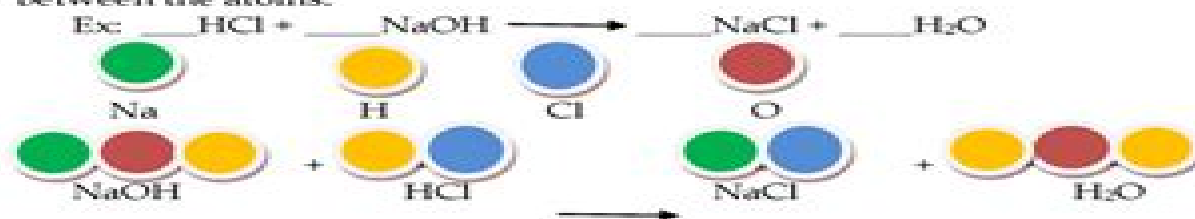
- Cup of Skittles
- 2 Paper Plates

### Procedure

1. Label one plate reactants and the other products, to represent each side of a chemical equation. Use at least four different colors of skittles (representing atoms of different elements) to make models of the substances in each equation below.



2. Choose one color to represent an element in the following chemical reaction.
3. To create a compound, put the correct color skittles together to represent a bond between the atoms.



4. Count the number of atoms you have total for your products and for your reactants, before you attempt to balance the equation. Look for the same colors on the products and reactants.

| Reactants |                  |                 | Products |                  |                 |
|-----------|------------------|-----------------|----------|------------------|-----------------|
|           | Before Balancing | After Balancing |          | Before Balancing | After Balancing |
| Na        | 1                |                 | Na       | 1                |                 |
| O         | 1                |                 | O        | 1                |                 |
| H         | 2                |                 | H        | 2                |                 |
| Cl        | 1                |                 | Cl       | 1                |                 |

# Balancing Equations Using Models Lab Answers

**William Collins**



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    **Energy Abstracts for Policy Analysis** ,1982    *SERL Report* University of California, Berkeley. Sanitary Engineering ResearchLaboratory,1967    **Report** National Physical Laboratory (Great Britain). Metrology Centre,1916    **Math Advantage** Grace M. Burton,Harcourt Brace,1998-05-22

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