

## Question 1

1(a)(iii) The three states of matter are solid, liquid and gas.

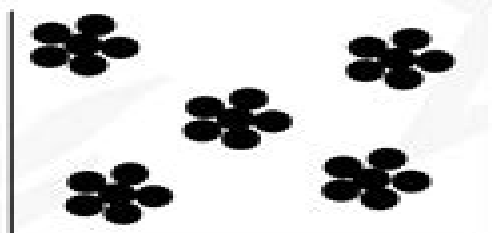
1(a)(iv) Melting point – the temperature at which a solid changes state to a liquid.

1(a)(v) The melting point of A is  $65^{\circ}\text{C}$ .

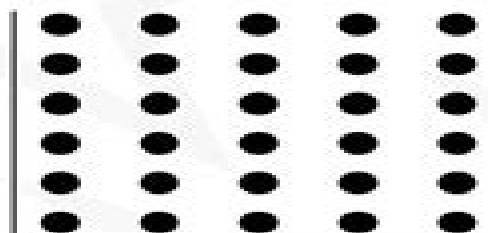
1(a)(vi) There would be a mixture of solid and liquid between  $t = 2.8$  minutes and  $t = 5.4$  minutes.

1(a)(vii) One precaution which should be taken during this experiment is that the boiling tube should be handled with a test tube holder, and not with bare/gloved hands.

1(a)(viii)



Particles of A at 0 min



Particles of A at 10 min

1(b)

	Inference
(i)	<ul style="list-style-type: none"> <li><math>\text{NO}_2</math> gas produced; <math>\text{NO}_3^-</math> present.</li> </ul>
(iii)	<ul style="list-style-type: none"> <li><math>\text{Ca}^{2+}_{(\text{aq})}</math>, <math>\text{Al}^{3+}_{(\text{aq})}</math>, <math>\text{Pb}^{2+}_{(\text{aq})}</math> and/or <math>\text{Zn}^{2+}_{(\text{aq})}</math> possibly present.</li> <li><math>\text{Al}^{3+}_{(\text{aq})}</math>, <math>\text{Pb}^{2+}_{(\text{aq})}</math> and/or <math>\text{Zn}^{2+}_{(\text{aq})}</math> possibly present.</li> </ul>
(iv)	<ul style="list-style-type: none"> <li><math>\text{Al}^{3+}_{(\text{aq})}</math>, <math>\text{Pb}^{2+}_{(\text{aq})}</math> and/or <math>\text{Zn}^{2+}_{(\text{aq})}</math> possibly present.</li> <li><math>\text{Al}^{3+}_{(\text{aq})}</math> or <math>\text{Pb}^{2+}_{(\text{aq})}</math> possibly present.</li> </ul>
(v)	<ul style="list-style-type: none"> <li><math>\text{Pb}^{2+}_{(\text{aq})}</math> confirmed to be present.</li> <li><math>\text{Pb}^{2+}_{(\text{aq})} + 2\text{I}^{-}_{(\text{aq})} \rightarrow \text{PbI}_{2(\text{s})}</math>.</li> </ul>

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