

**Covalent Bonding Quiz (24 Points)****Section I – Basic Concepts & Vocabulary (12 points – 1 pt. each)**

**Instructions:** For Items 1–4, circle the correct response.

1. A *covalent bond* is formed by \_\_\_\_\_ electrons between two nonmetal atoms.  
a) freezing                      c) sharing  
b) transferring                d) melting
2. A group of nonmetal atoms, joined by *covalent bonds* is defined as a(an)  
a) molecule                      c) metal  
b) salt crystal                  d) octet
3. A *structural formula* uses Lewis dot structures to show how atoms are bonded, including the number & types of bonds, in a covalent molecule? (Yes or No)
4. Do atoms, except hydrogen, in covalent bonding follow the octet rule? (Yes or No)

**Instructions:** Match the word on the left with the definition on the right by writing the letter on the line next to the word that is a “definition match” for that word.

- |                            |       |  |       |  |
|----------------------------|-------|--|-------|--|
| 5. molecular formula       | _____ | A. a covalent bond in which only one pair of electrons is shared by two bonded atoms   |       |  |
|                            |       | 6. nonpolar covalent bond  | _____ | B. chemical formula that shows the arrangement of atoms in molecules and polyatomic ions |
| 7. triple covalent bond    | _____ | C. a covalent bond between two atoms of different electronegativities in which the bonding electrons are not shared equally      |       |  |
| 8. double covalent bond    | _____ | D. a covalent bond formed by the equal sharing of bonding electrons by two atoms   |       |  |
| 9. polar covalent bond     | _____ | E. a covalent bond in which two pairs of electrons are shared by two bonded atom   |       |  |
| 10. single covalent bond   | _____ | F. two valence electrons that are <u>not involved in bonding</u> at a bond site in a covalent molecule                           |       |  |
| 11. structural formula     | _____ | G. indicates the exact number of atoms in a covalent compound, but does NOT show the arrangement of the atoms nor types of bonds |       |  |
| 12. unshared electron pair | _____ | H. a covalent bond in which three pairs of electrons are shared by two bonded atom   |       |  |

# Covalent Bonding Quiz With Answers Student Edgenuity

**Jicheng Xie**



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